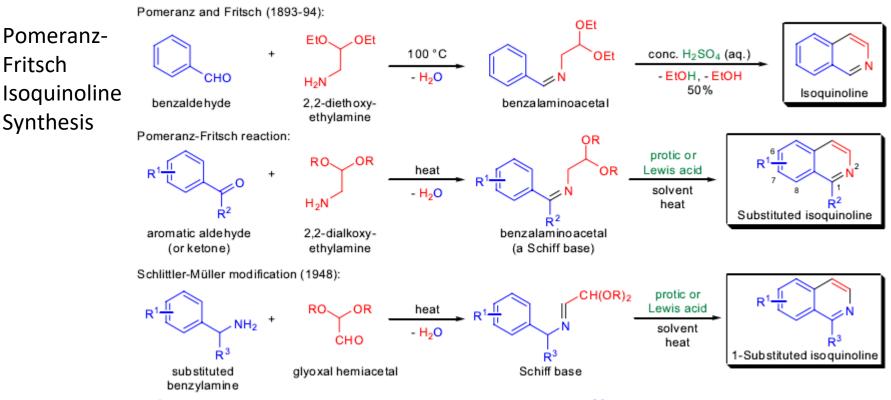
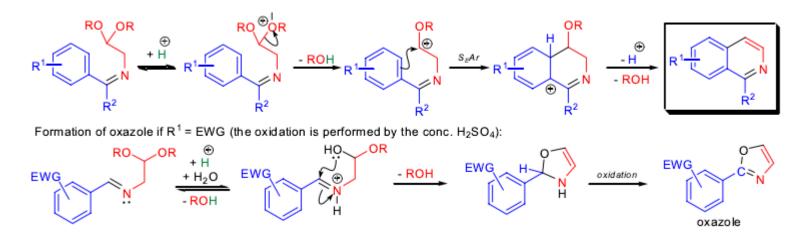
Isoquinolines Synthesis

Liangleiming 20160103



R<sup>1</sup> = usually an electron-donating group (EDG), H, alkyl, aryl, O-alkyl, Cl, Br; R<sup>2-3</sup> = H, alkyl; R = Me, Et; <u>protic acid</u>: H<sub>2</sub>SO<sub>4</sub>, HCl, PPA; <u>Lewis acid</u>: BF<sub>3</sub>·OEt<sub>2</sub>

#### Mechanism:



features : 1) the benzalaminoacetals are prepared by reacting 2,2-dialkoxyethylamines with substituted aromatic aldehydes or rarely with aromatic ketones;

2) the structural variation of the 2,2-dialkoxyethylamines is very restricted, and, in the overwhelming majority of the cases, the dimethyl or diethyl acetals are used without any substituents on the C1 carbon (C1-substituted analogues tend to fail to undergo the reaction);

3) aromatic aldehydes give rise to C1-unsubstituted isoquinolines, usually in good yield, while aromatic ketones afford C1-substituted isoquinolines albeit in low yield;

4) the highest yields are obtained when the substituents on the aromatic ring are electron-donating; 5) strongly electron-withdrawing substituents (e.g., NO2) on the aromatic ring prevent the formation of isoquinolines and the corresponding oxazoles are obtained instead;

6) when both of the ortho-positions (relative to the carbonyl group) are unoccupied, a regioisomeric mixture of isoquinolines is obtained;

7) the most commonly used protic acids are sulfuric acid and hydrochloric acid, but Lewis acids such as BF3·OEt2, trifluoroacetic anhydride and lanthanide triflates have been used occasionally;
8) unless the aromatic ring is highly electron-rich, heating of the reaction

mixture is required in order to achieve cyclization.

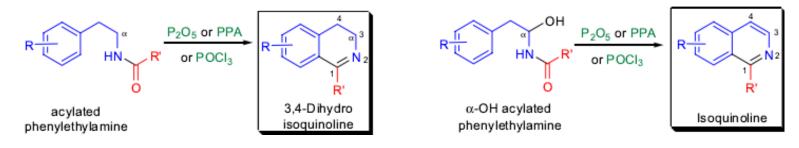
Two of the most important modifications are:

1) when a substituted benzylamine is condensed with glyoxal hemiacetal, the resulting Schiff base is efficiently cyclized to give the corresponding C1-substituted isoquinoline (Schlittler-Müller modification);

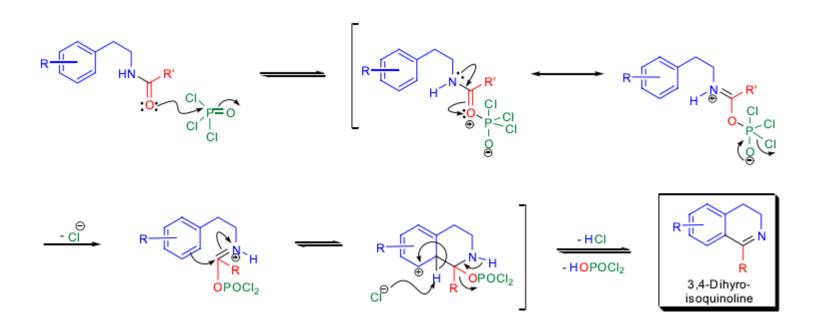
2) hydrogenation of the benzal-aminoacetal and the acid-catalyzed cyclization of the resulting amine gives rise to a tetrahydroisoquinoline (Bobbitt-modification).

## Bischler-Napieralski Isoquinoline Synthesis

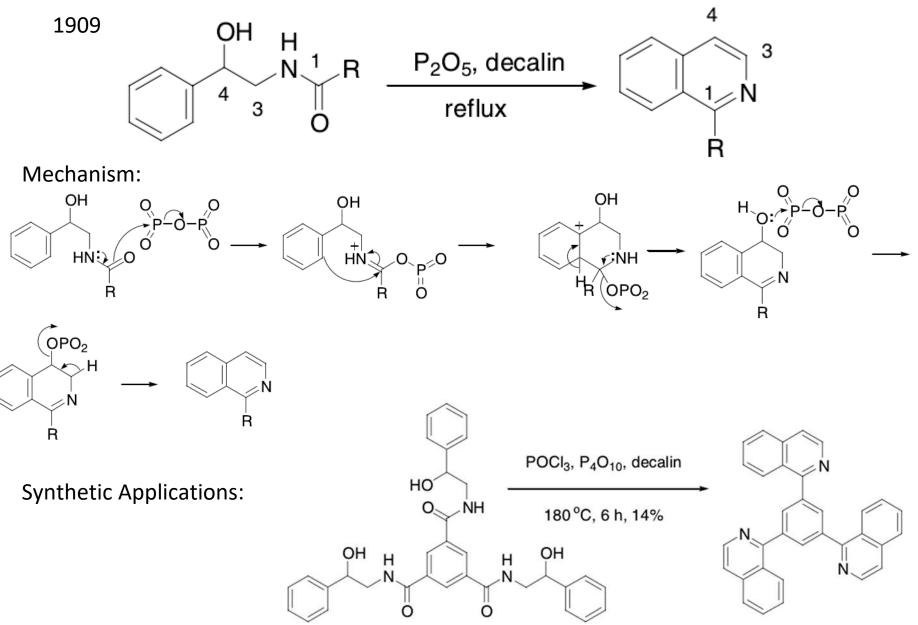
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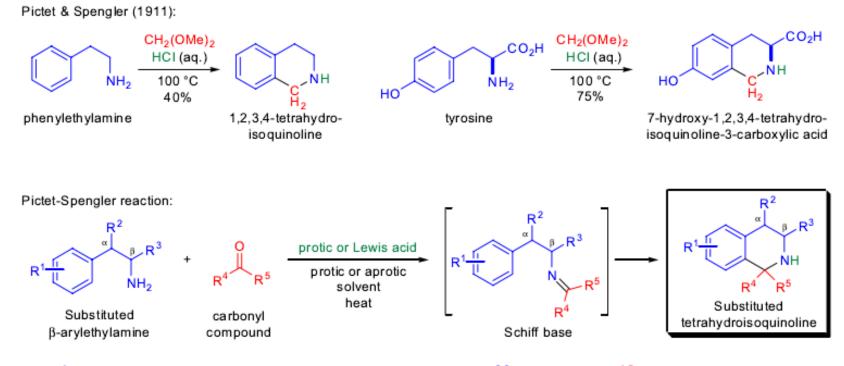
Mechanism:



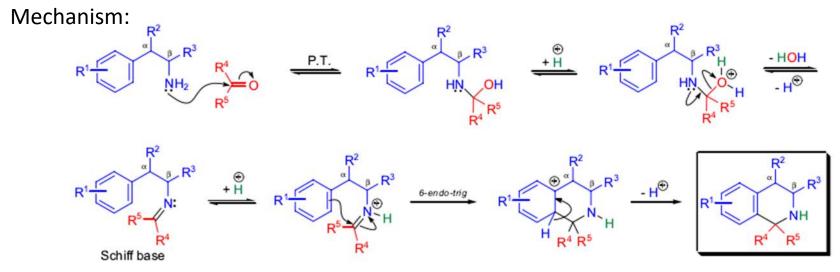
### Pictet–Gams isoquinoline synthesis



### Pictet-spengler Tetrahydroisoquinoline Synthesis



 $R^{1} = H$ , alkyl, aryl, O-alkyl, usually an electron-donating group (EDG);  $R^{2-3} = H$ , alkyl, aryl;  $R^{4-5} = H$ , alkyl, aryl; <u>protic acid</u>: HCl, H<sub>2</sub>SO<sub>4</sub>, TFA, silica gel; <u>Lewis acid</u>: BF<sub>3</sub>-OEt<sub>2</sub>



# features:

- 1) only  $\beta$ -arylethylamines with electron-donating substituents afford high yields;
- 2) the carbonyl compound can be an aldehyde or a ketone or any acid-labile surrogate;
- 3) the most frequently used aldehyde is formaldehyde or its dimethyl acetal;
- 4) the number of electron-donating groups on the aromatic ring influences the ease of the reaction, and, for example, the presence of two alkoxy groups allows the Pictet-Spengler reaction to proceed under physiological conditions (this is important in the biosynthesis of alkaloids);
- 5) the reaction is usually carried out with a slight excess of the carbonyl compound (to ensure the complete consumption of the amine) in either protic or aprotic medium;

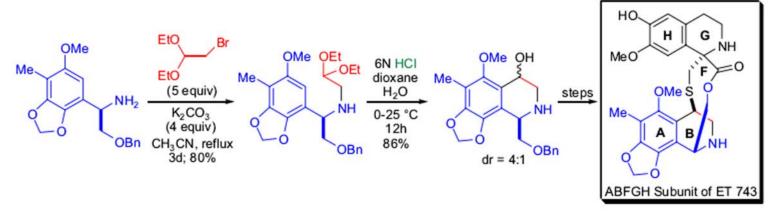
6) since the reaction goes through the intermediacy of a Schiff base, the Schiff base can be prepared separately and subjected to a protic or Lewis acid to afford the cyclized tetrahydroisoquinoline product.

## Synthetic Applications:

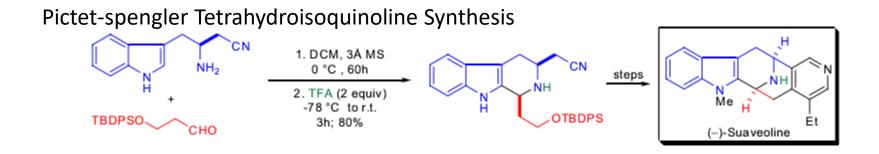


Pictet-spengler Tetrahydroisoquinoline

Bobbitt modified Pomeranz-Fritsch Isoquinoline Synthesis



Zhou, B., Guo, J., Danishefsky, S. J. Studies Directed to the Total Synthesis of ET 743 and Analogues Thereof: An Expeditious Route to the ABFGH Subunit. Org. Lett. 2002, 4, 43-46.



#### Bischler-Napieralski Isoquinoline Synthesis

