NOZAKI-HIYAMA-KISHI REACTION

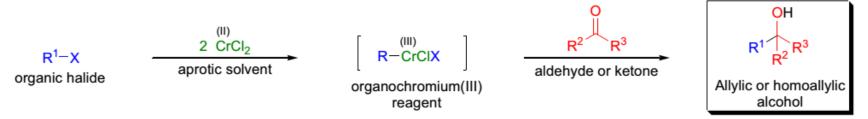
Origin

- In 1977, H. Nozaki and T. Hiyama et al. reacted aldehydes and ketones with organochromium(III) reagents, which were generated in situ from allyl and vinyl halides upon treatment with CrCl2 under aprotic and oxygen-free conditions, and obtained the corresponding allylic and homoallylic alcohols with high chemospecificity and stereoselectivity.
- In 1986, Y. Kishi and H. Nozaki independently discovered that traces of nickel salts catalyzed the formation of carbon-chromium(III) bonds, even from otherwise less reactive substrates (e.g., vinyl and aryl halides). This modification helped to make the process more reliable.
- The one-pot addition of alkenyl, alkynyl, aryl, allyl, or vinylchromium compounds to aldehydes or ketones is known as the Nozaki-Hiyama-Kishi (NHK) reaction.

$$\begin{array}{c} R^{1-X} \\ \text{organic halide} \end{array} \begin{array}{c} \underbrace{ \begin{array}{c} \text{(II)} \\ 2 \text{ CrCl}_2 \\ \text{aprotic solvent} \end{array}}_{\text{organochromium(III)}} \\ \underbrace{ \begin{array}{c} \text{R}^{1-X} \\ \text{aldehyde or ketone} \end{array}}_{\text{aldehyde or ketone}} \\ \underbrace{ \begin{array}{c} \text{OH} \\ \text{R}^{1} \\ \text{R}^{2} \\ \text{Allylic or homoallylic alcohol} \end{array}}_{\text{alcohol}}$$

R¹ = alkenyl, aryl, allyl, vinyl, propargyl, alkynyl, allenyl; X = Cl, Br, I, OTf, etc.; R², R³ = alkyl, aryl, alkenyl, H; solvent: DMF, DMSO, THF

Features&Drawbacks



R¹ = alkenyl, aryl, allyl, vinyl, propargyl, alkynyl, allenyl; X = Cl, Br, I, OTf, etc.; R², R³ = alkyl, aryl, alkenyl, H; solvent: DMF, DMSO, THF

Features:

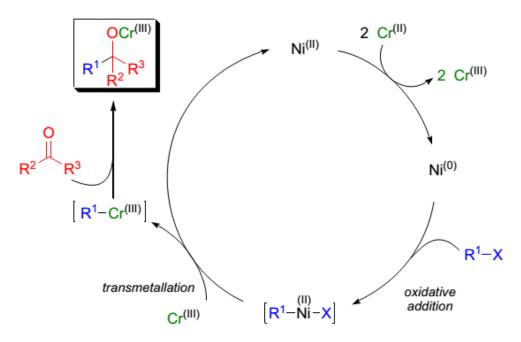
- ➤ The CrCl2 is either purchased commercially or prepared by the reduction of CrCl3 prior to the reaction.
- Cr(II) is a one-electron donor, and therefore two moles of the chromium(II) salt are required to reduce one mol of organic halide to the corresponding organochromium(III) reagent.
- aldehydes react markedly faster than ketones.
- because of their low basicity, organochromium reagents are compatible with a wide range of sensitive functional groups.

Drawbacks:

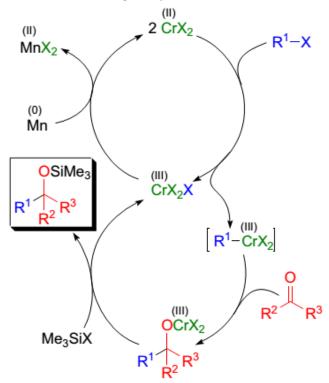
- The nickel and chromium salts are very toxic.
- Usually a large excess of CrCl2 is required.
- The Lewis acidic salts formed during the preparation of CrCl2 may alter the stereochemical outcome of the reaction for polyfunctional substrates where chelation control determines the stereochemical course.

Mechanism

Ni^(II)-catalyzed process:



Chromium-catalyzed process:



Synthetic Applications

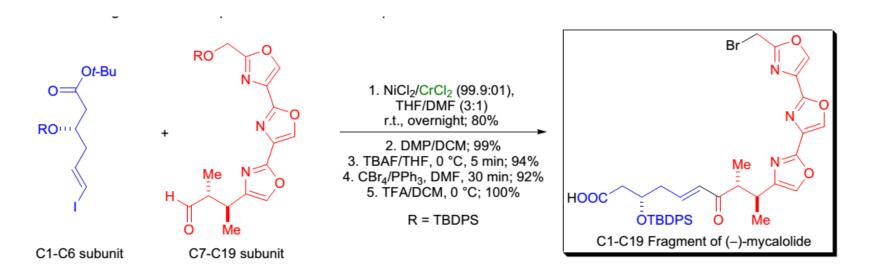
CrCl₂ (8.7 equiv) NiCl₂ (1 mol%)

DMF/THF r.t., 12h; 88%

$$\alpha:\beta = 2:1$$

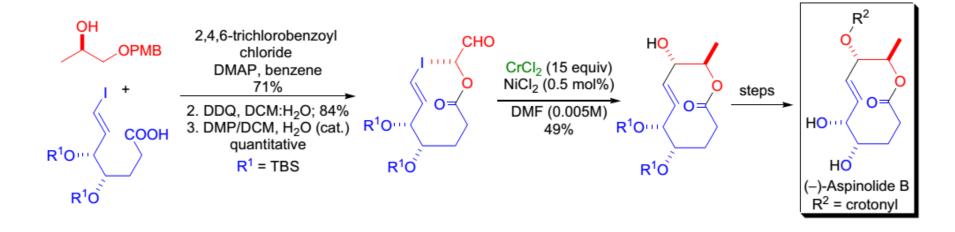
OAc

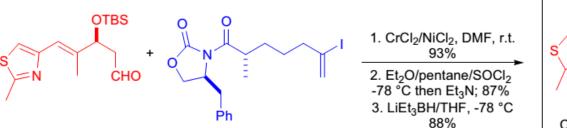
Deacetoxyalcyonin Acetate

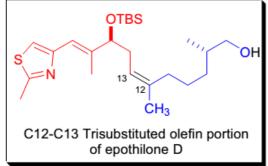


Panek, J. S., Liu, P. Total Synthesis of the Actin-Depolymerizing Agent (-)-Mycalolide A. *J. Am. Chem. Soc.* **2000**, 122, 11090-11097.

Synthetic Applications







Taylor, R. E., Chen, Y. Total Synthesis of Epothilones B and D. Org. Lett. 2001, 3, 2221-2224.

Thanks