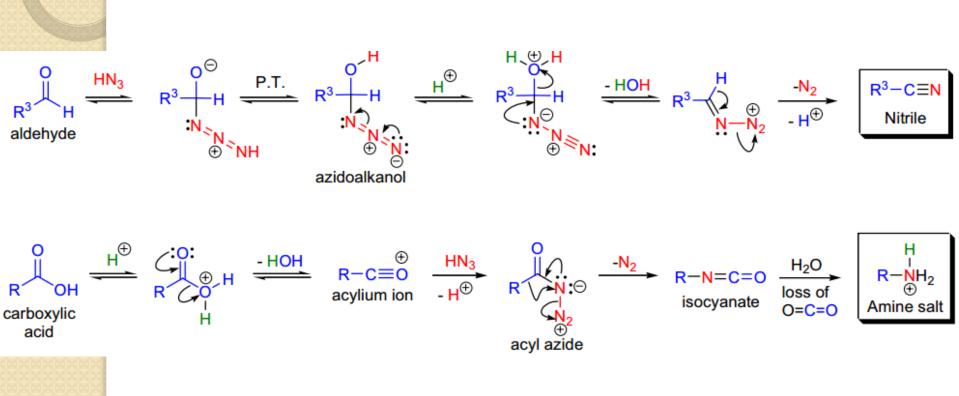


 R^{1-2} = alkyl, aryl; Ar = substituted aryl or heteroaromatic; R^3 = Me, substituted aryl; R^4 = alkyl, substituted alkyl; R^5 = aryl; R^6 = H, alkyl or aryl; <u>Lewis acid</u>: TiCl₄, TFA, CH₃SO₃H; <u>acid catalyst</u>: H₂SO₄, PPA, Cl₃CCO₂H/H₂SO₄, TFA, TFAA



Mechanism

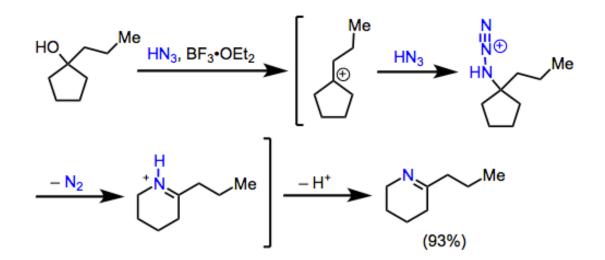


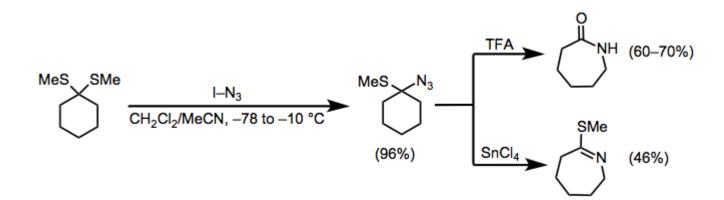
General features

- Hydrogen azide is handled either as a solution in an inert solvent (e.g., CHCl₃) or generated in situ by adding NaN₃ to the acidic reaction mixture; HN₃ is known to be toxic and explosive.
- 2) The product amines are one-carbon shorter homologs of the substrates due the loss of CO₂
- 3) Aromatic acids with electron-withdrawing groups require the use of very strong acid catalysts (e.g., conc. H₂SO₄ or oleum)
- 4) The α-stereocenter remains unaffected and the product amine is obtained with retention of configuration;
- 5) 1,3-dicarboxylic acids react at only one of the carboxylic acid fuctional groups; α-amino acids do not react.
- Aldehydes and ketones react with hydrazoic acid faster than carboxylic acids so good chemoselectivity can be achieved with keto acids;

Reactions of Other Electrophiles

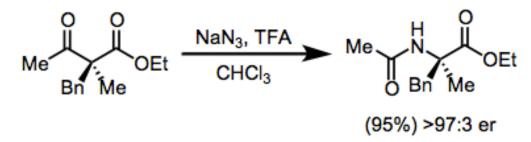
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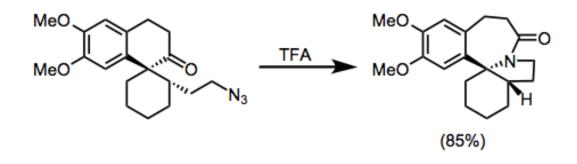


Synthetic Applications

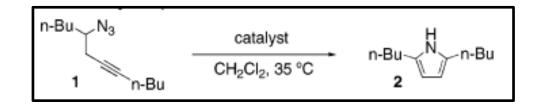
The Schmidt reaction has been applied to the synthesis of disubstituted amino ^oacids. After asymmetric alkylation of a β-ketoester, treatment with trifluoroacetic acid (TFA) and sodium azide yields the N-acetyl amino ester, which may be converted to the corresponding amino acid through functional group manipulations.



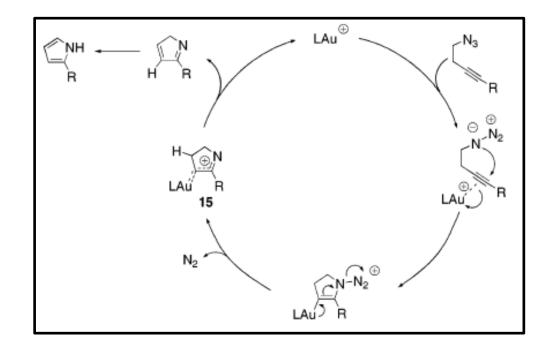
The Schmidt reaction has also been applied extensively to access crowded or otherwise difficult amines and amides in alkaloid natural products. Efforts toward the homoerythrina spirocyclic ring system contain a nice application of the intramolecular Schmidt reaction of alkyl azides. An appropriately positioned alkyl azide was smoothly converted to the lactam in the presence of TFA.



Gold(I)-Catalyzed Intramolecular Acetylenic Schmidt Reaction



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D. J. Gorin, N. R. Davis, F. D. Toste, J. Am. Chem. Soc., 2005, 127, 11260-11261.