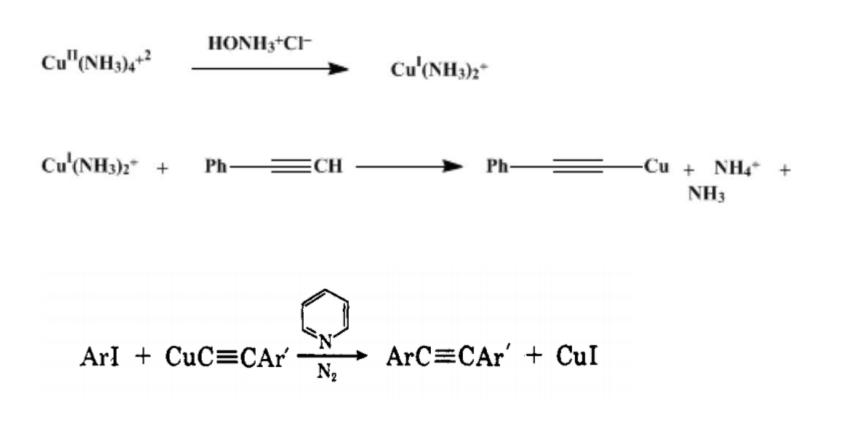
# **Sonogashira Cross-Coupling**



• In 1963, Castro-Stephens coupling



*J. Org. Chem.*, **1963**, *28* (12), pp 3313–3315 *Org. Synth.* 1972, *52*, 128 • In 1975, K. Sonogashira

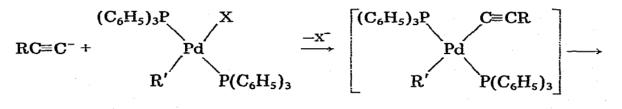
PhI + HCECH CuI-(Ph<sub>3</sub>P)<sub>2</sub>PdCl<sub>2</sub> PhCECPh Et<sub>2</sub>NH

K. Sonogashira Tetrahedron Lett. 1975, 4467-4470.

• In 1975, R.F. Heck and L. Cassar

 $ArX + RC \equiv CH + NaOCH_3 \xrightarrow{Pd[P(C_6H_5)_3]_4} ArC \equiv CR + NaX + CH_3OH$ 

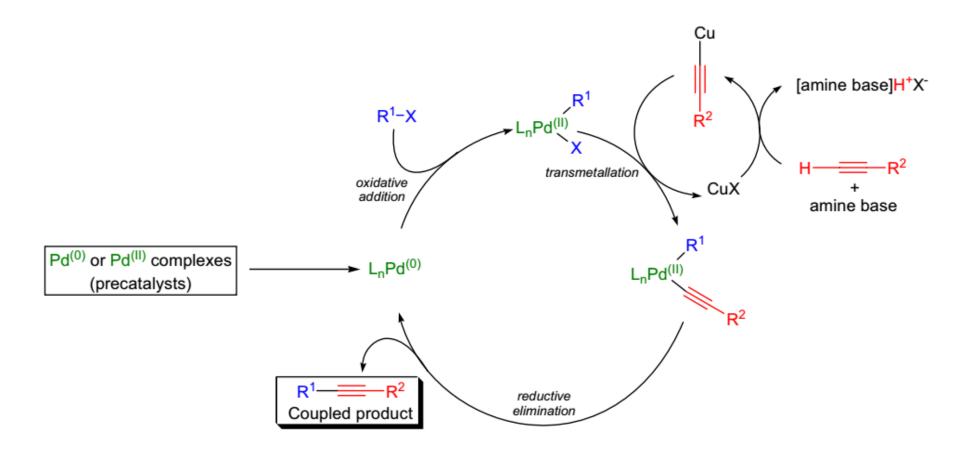
L. Cassar J. Organomet. Chem. 1975, 93, 253-257



 $\rightarrow \mathrm{R}'\mathrm{C} \equiv \mathrm{CR} + \mathrm{Pd}[\mathrm{P}(\mathrm{C}_{6}\mathrm{H}_{5})_{3}]_{2}$ 

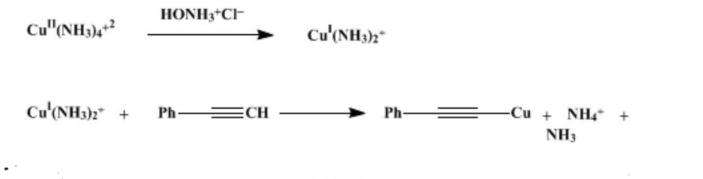
R.F. Heck J. Organomet. Chem. 1975,93, 259-263

## Mechanism:



Features of Sonogashira cross-coupling reaction:

- 1) The coupling can usually be conducted at or slightly above **room temperature**
- 2) The best palladium catalysts are Pd(PPh3)2Cl2 or Pd(PPh3)4
- 3) The handling of the shock-sensitive/explosive copper acetylides is avoided by the use of a catalytic amounts of copper(I) salt (Cul or CuBr, 0.5-5 mol%)



 $ArX + RC \equiv CH + NaOCH_3 \xrightarrow{Pd[P(C_6H_5)_3]_4} ArC \equiv CR + NaX + CH_3OH$ 

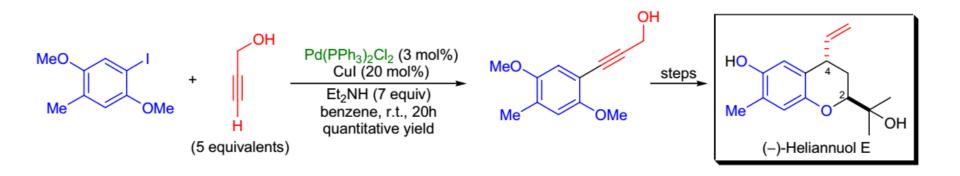
Features of Paterno-Büchi reaction:

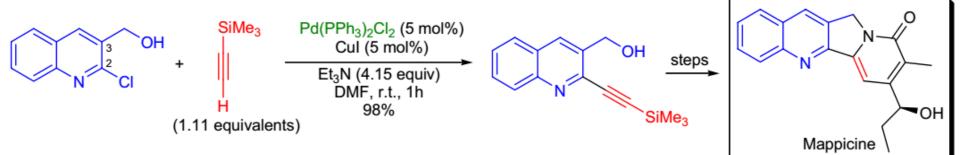
- 3) The solvents and the reagents **do not need to be rigorously dried**. However, a **thorough deoxygenation** is essential to maintain the activity of the Pd-catalyst.
- 4) When the copper co-catalyst is exposed to air, the copper acetylide undergoes homocoupling with itself instead of undergoing the desired cross-coupling with the vinyl or aryl halide compound
- 5) **Stereochemical** information of the substrates is **preserved** in the products;
- 6) The order of reactivity for the aryl and vinyl halides is  $I \approx OTf > Br >> Cl$ ;
- 7) Alkynes with conjugated EWG(R2=CO2Me) give Michael addition products ; (R2=CH2CO2Me or NH2, tend to rearrange to allenes)  $\sim C \sim R_2$
- 8) Aryl halides and bulky substrates that are not very reactive require higher reaction temperature and at high temperatures terminal akynes undergo side reactions.

Efforts to Modify Reaction Conditions:

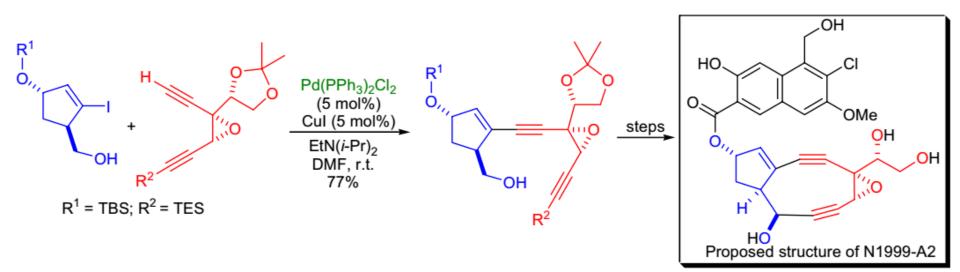
- 1. The original Palladium catalyst, (Ph3P)2PdCl2, has been occasionally substituted with NaPdCl4, Pd(OAc)2, Pd(MeCN)2Cl2, PdCl2, and PdI2 catalysts.
- 2. The copper co-catalyst has been replaced with other zinc, tin, boron, aluminum.

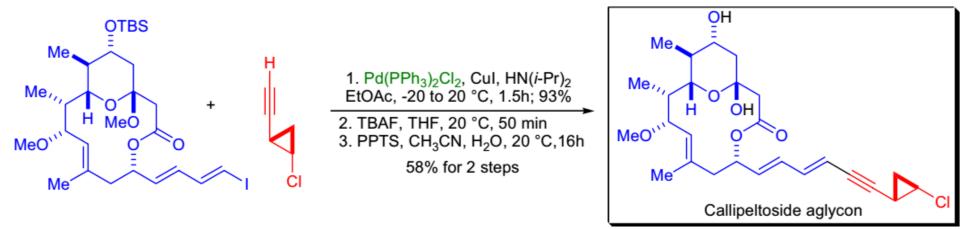
### Applications:



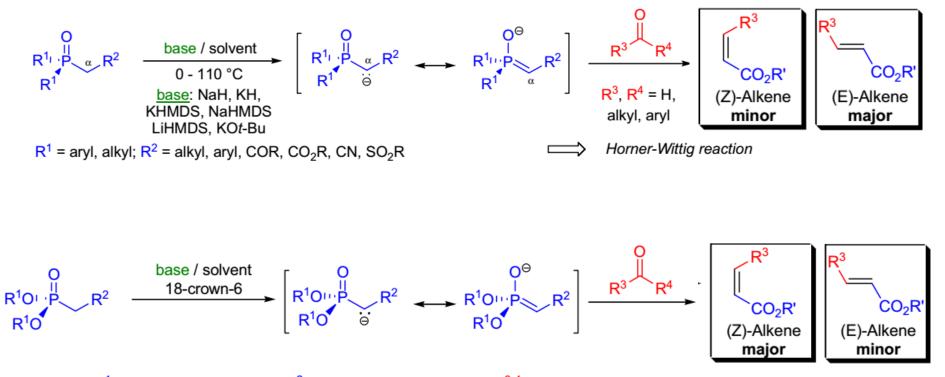


### Applications:





#### 1. Wittig reaction , HWE olefination



 $R^1 = CH_2CF_3$ , trifluoroalkyl;  $R^2 = COR$ ,  $CO_2R$ , CN,  $SO_2R$ ;  $R^{3-4} = H$ , alkyl, aryl; <u>base</u> = KH, KHMDS