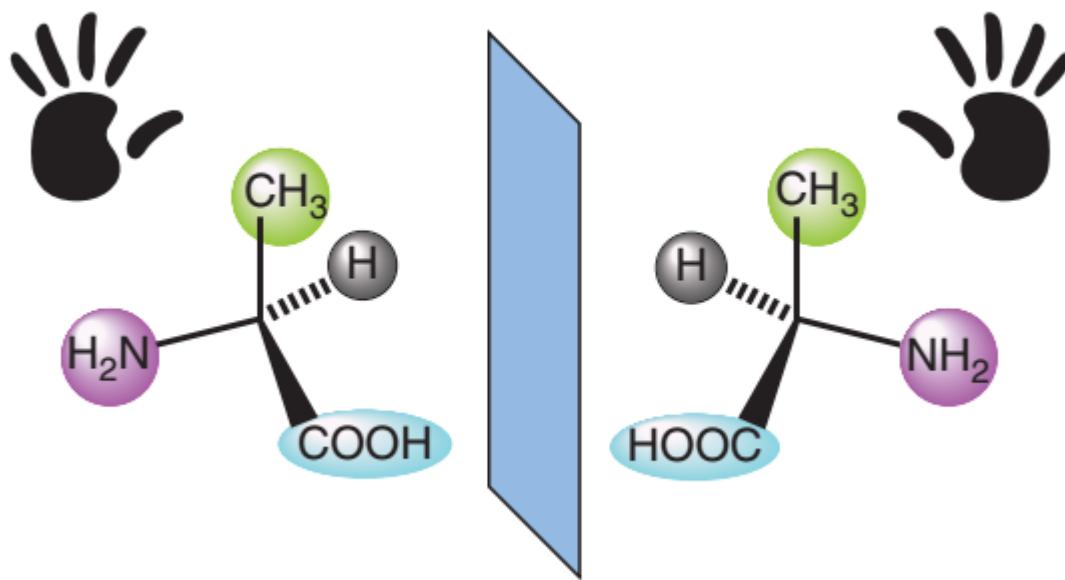
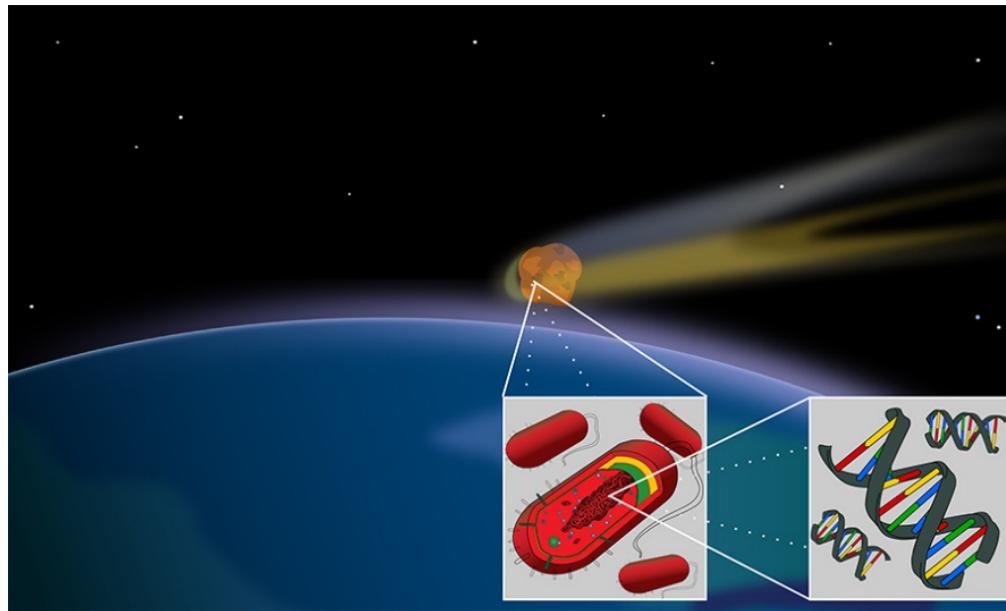


The Origin of Biological Homochirality



Origin of Life Hypothesis

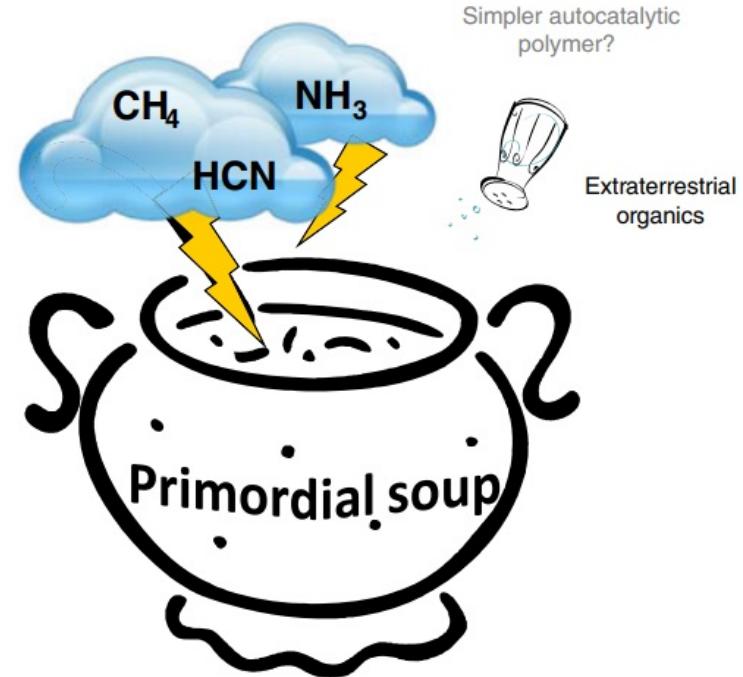
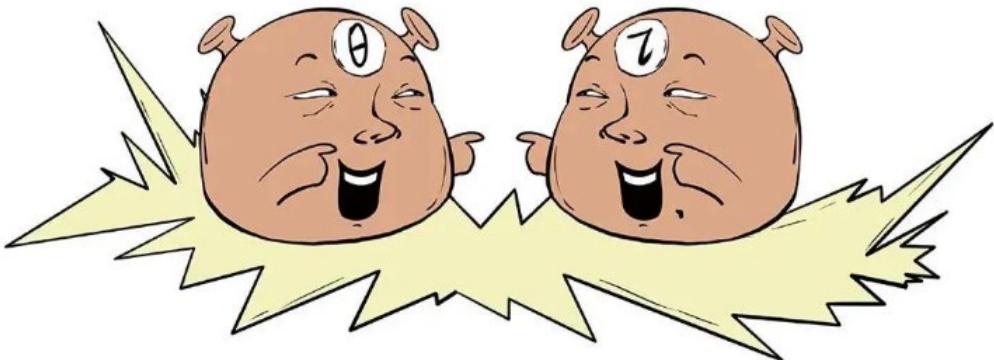


这么眼熟，兄dei

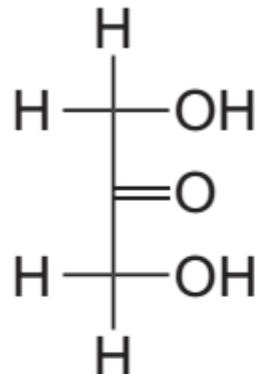
“

兄dei嘛，当然眼熟

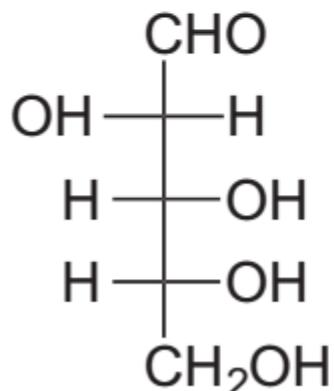
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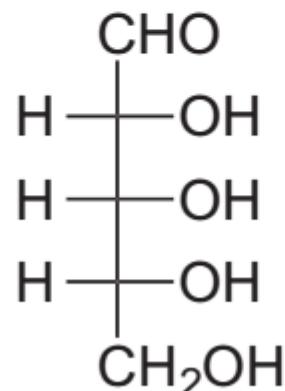
Panspermia Hypothesis



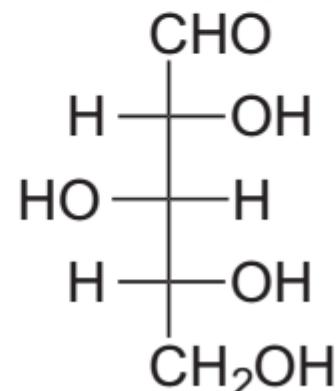
Dihydroxyacetone



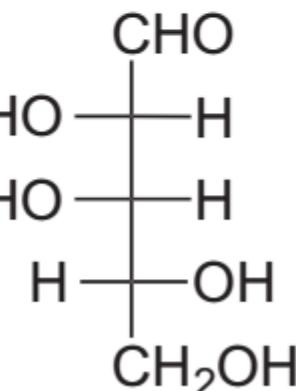
D-arabinose



D-ribose

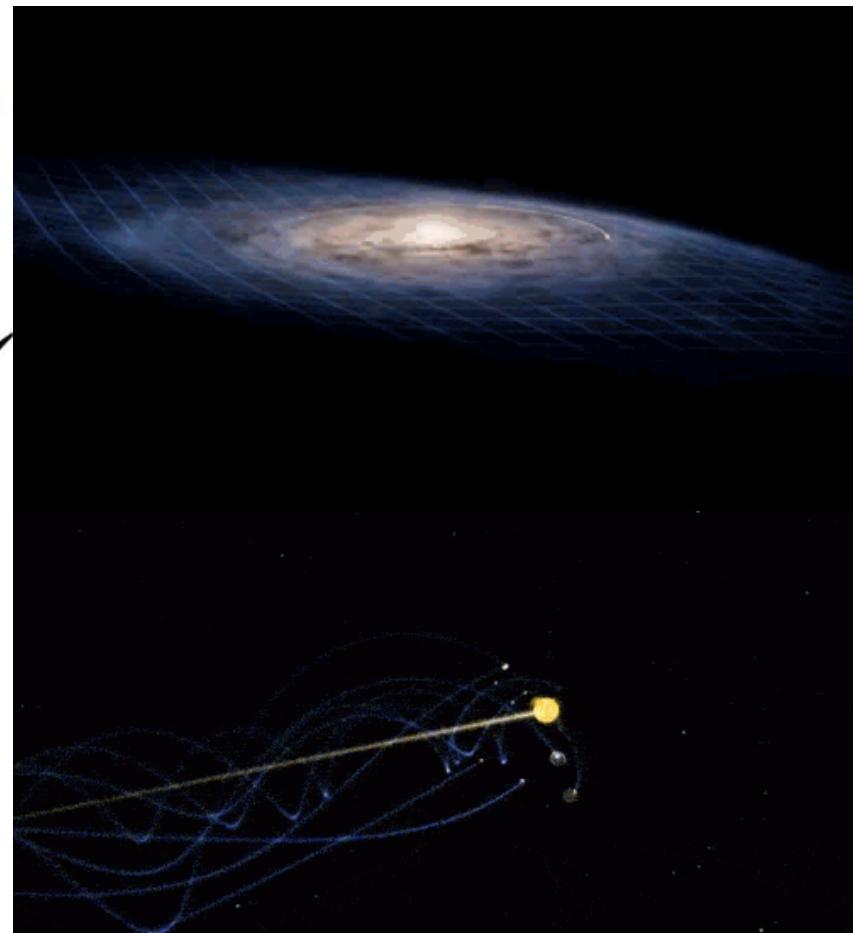
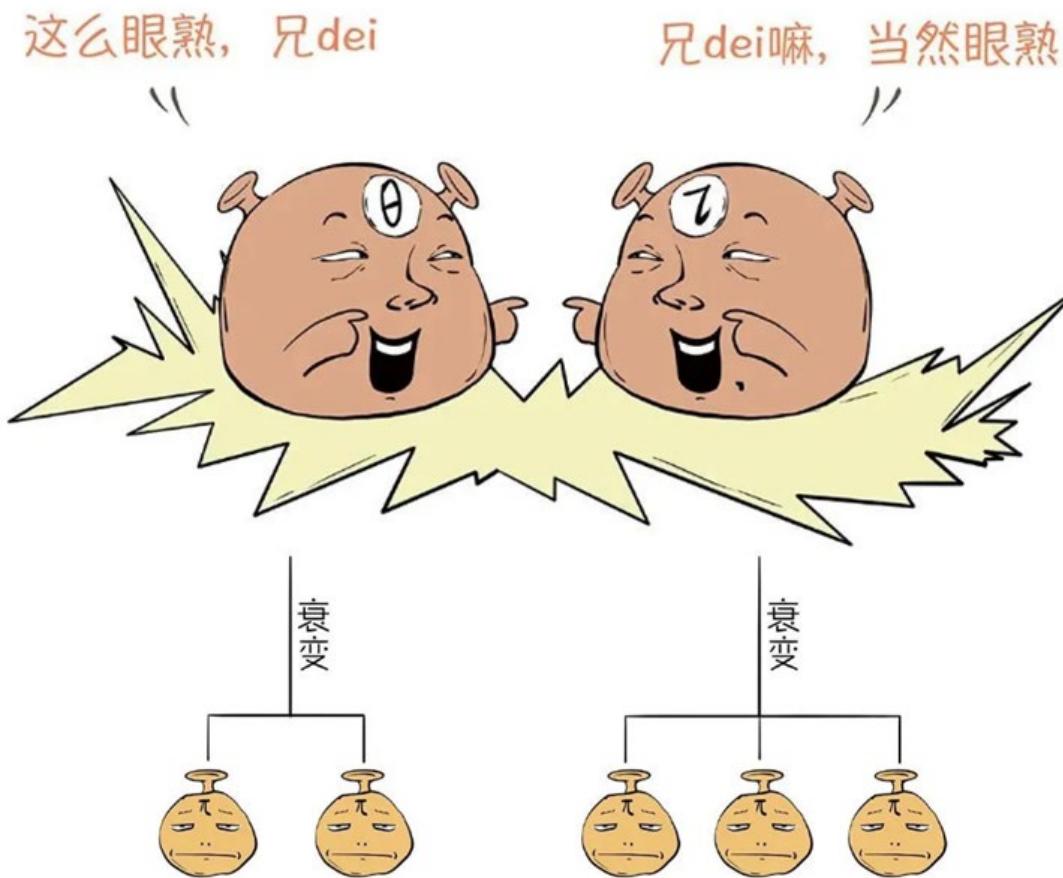


D-xylose



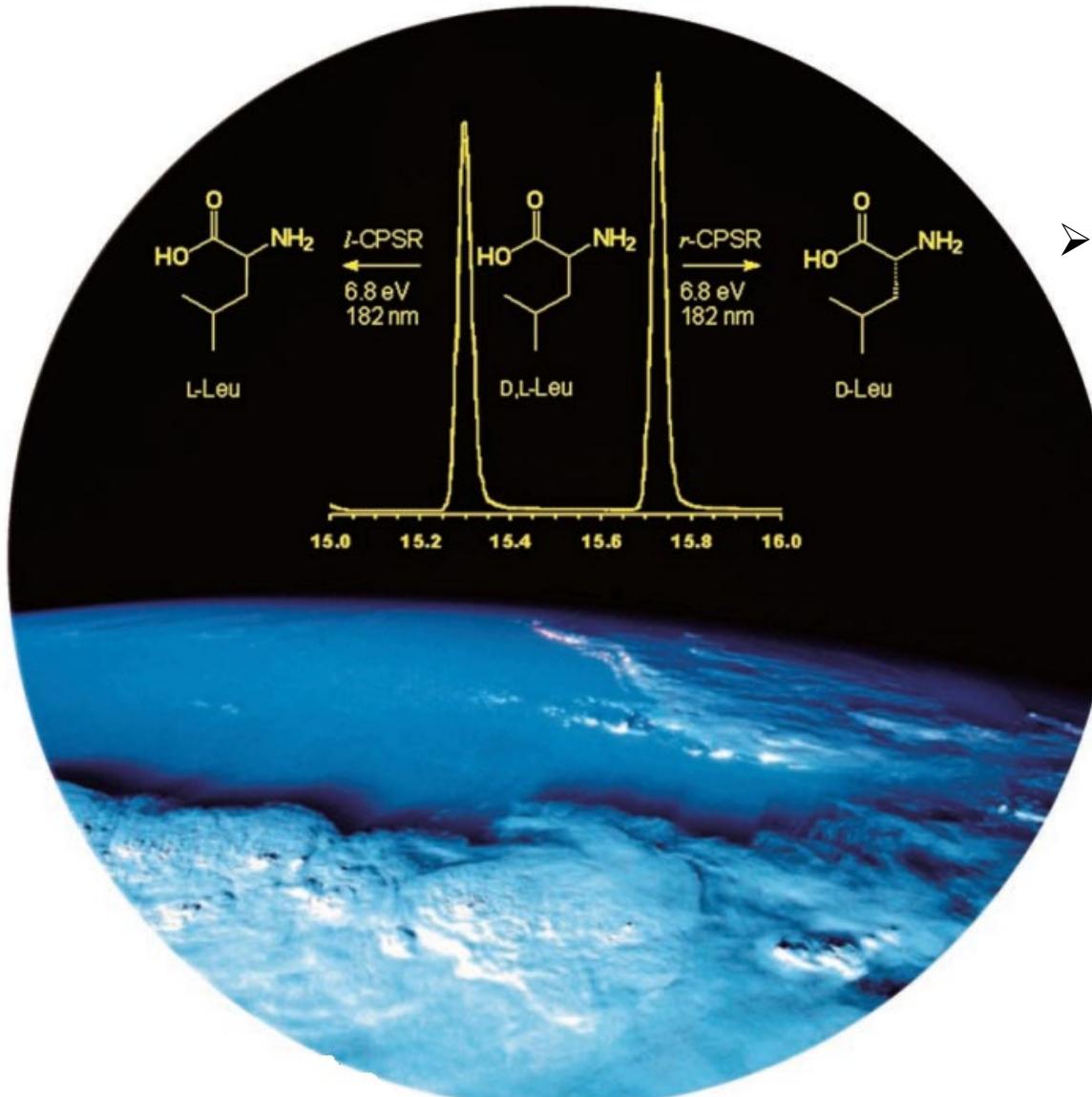
D-lyxose

Parity Violation



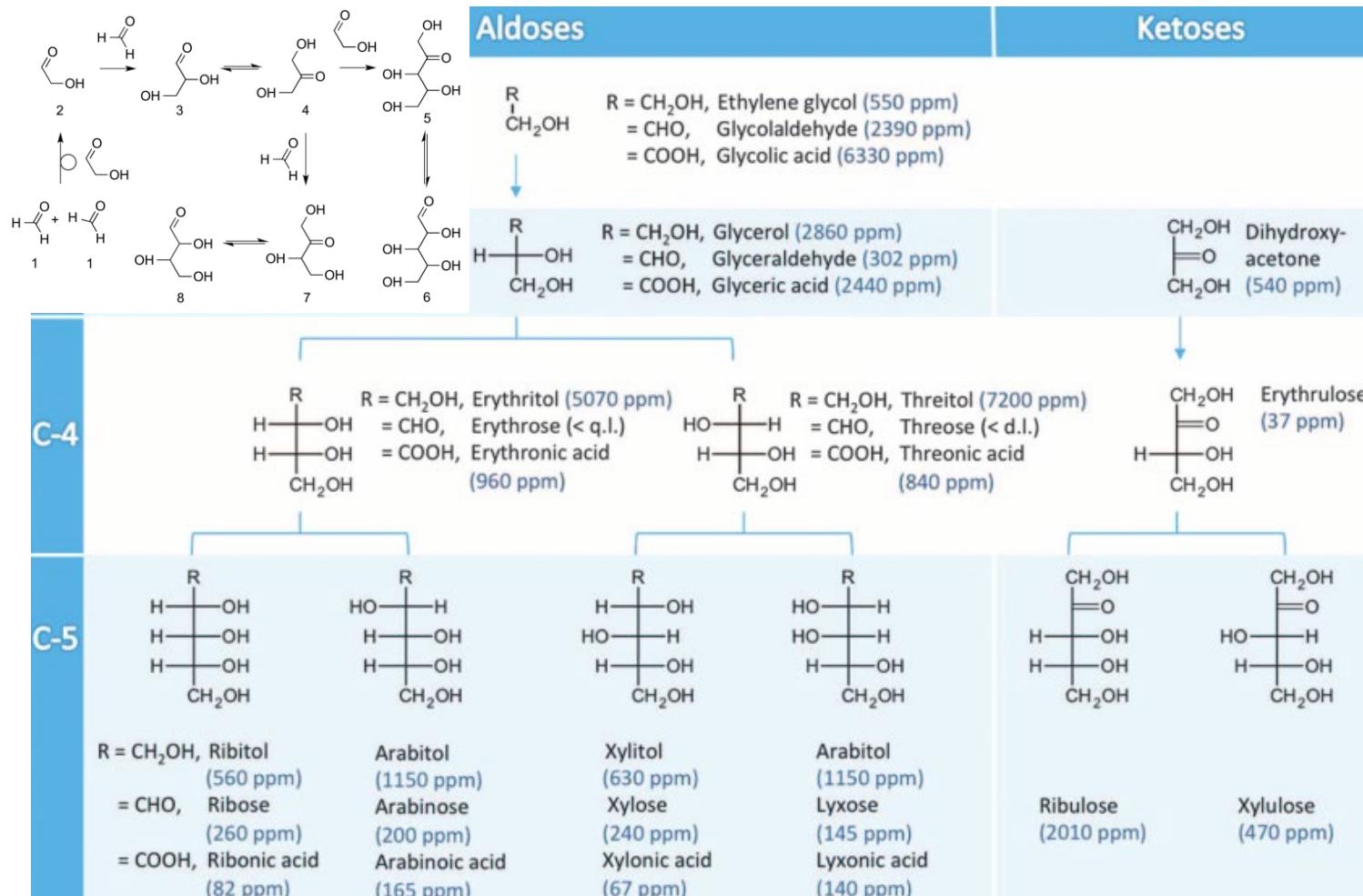
1. Cosmic dust particles to polarize starlight as circularly polarized in one direction only.
2. The weak nuclear force (radioactive decay) is chiral. During beta decay, the emitted electrons preferentially favor one kind of spin.

Panspermia Hypothesis-Parity Violation



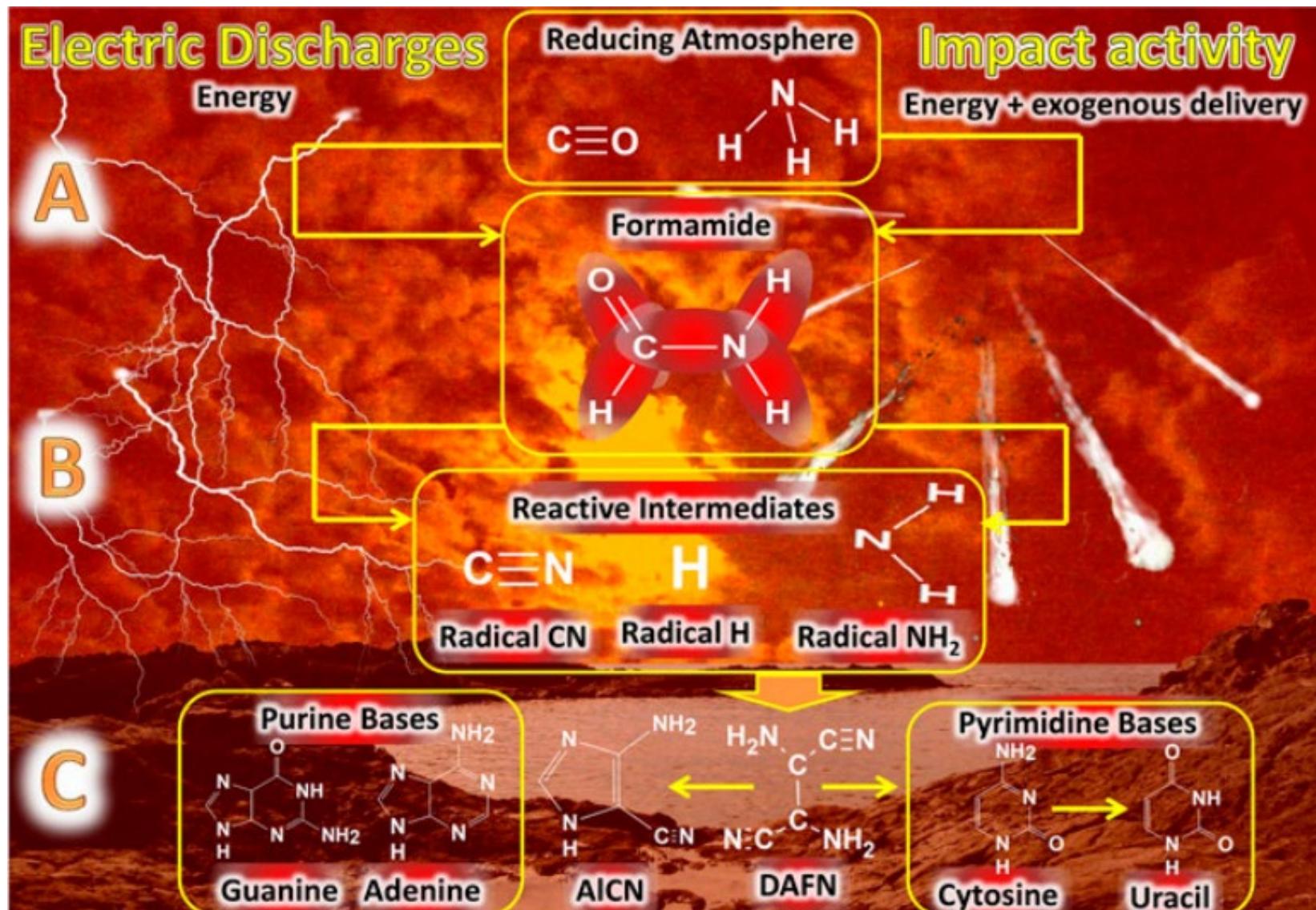
- This circularly polarized light degrades D- amino acids more than L enantiomers, and this effect is clear when analyzing the amino acids found on comets and meteors.

Parity Violation

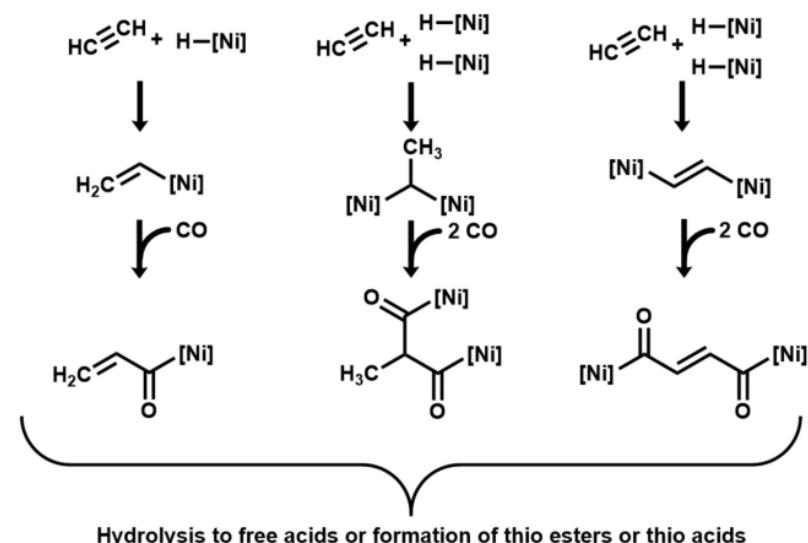
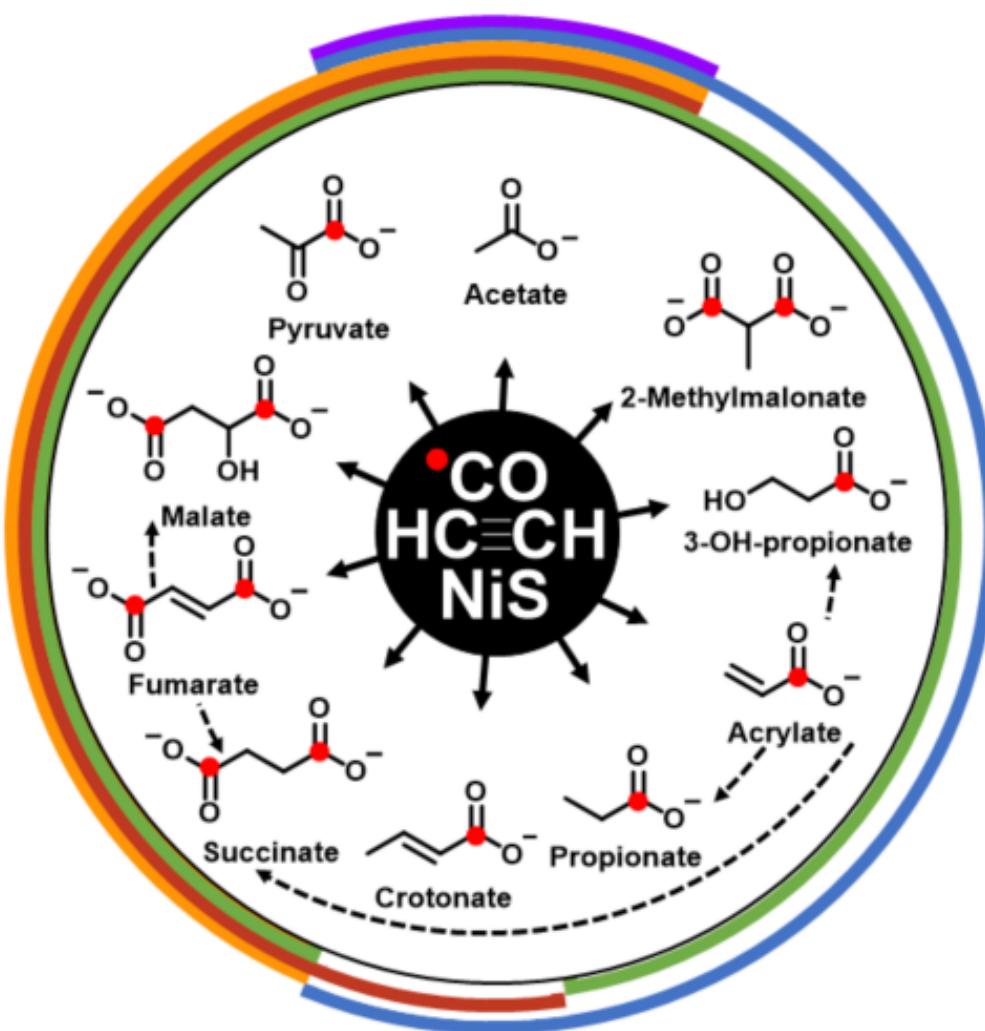


Nature, 2002, 416, 401
Science, 2016, 352, 208

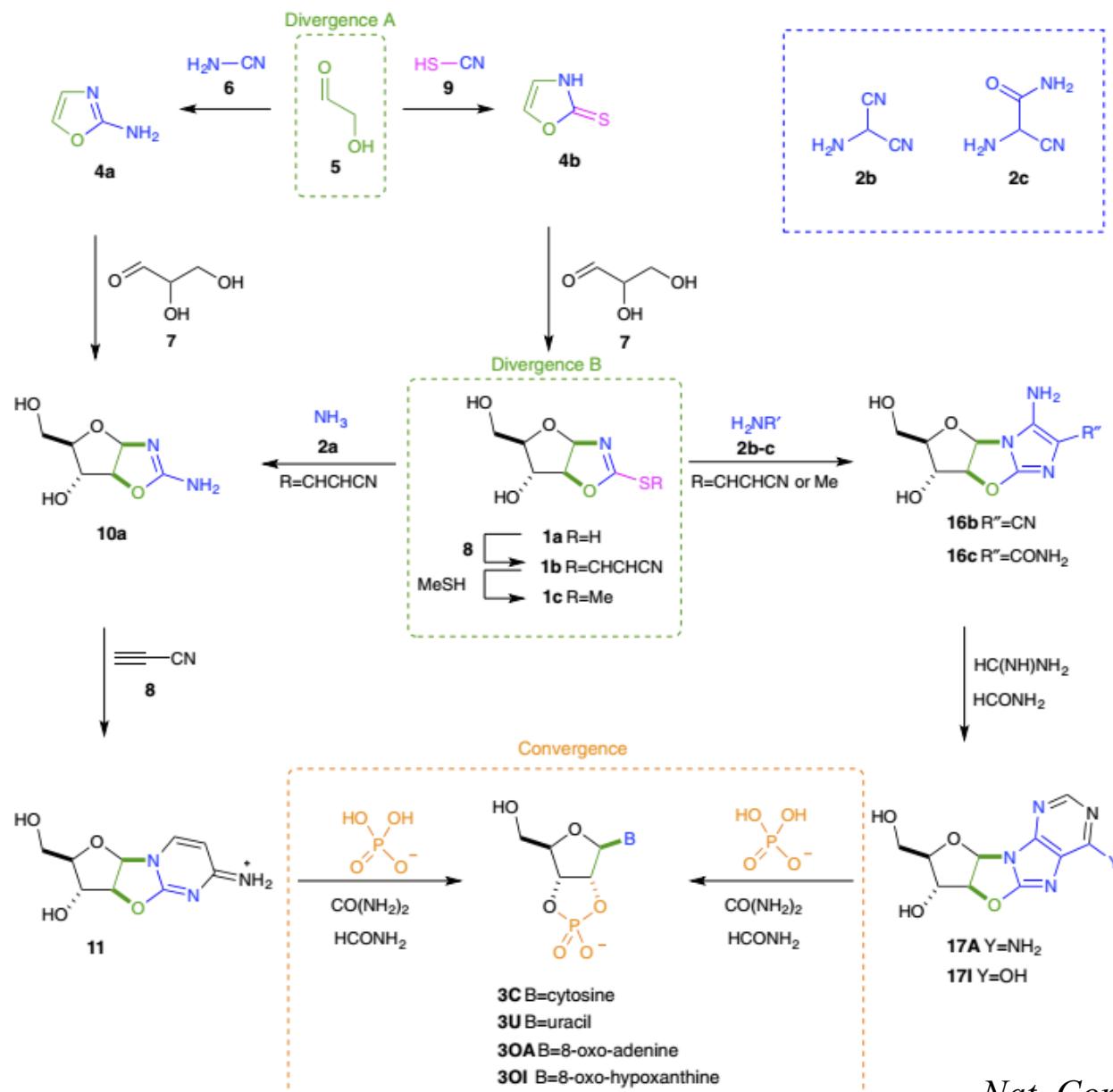
Prebiotic soup



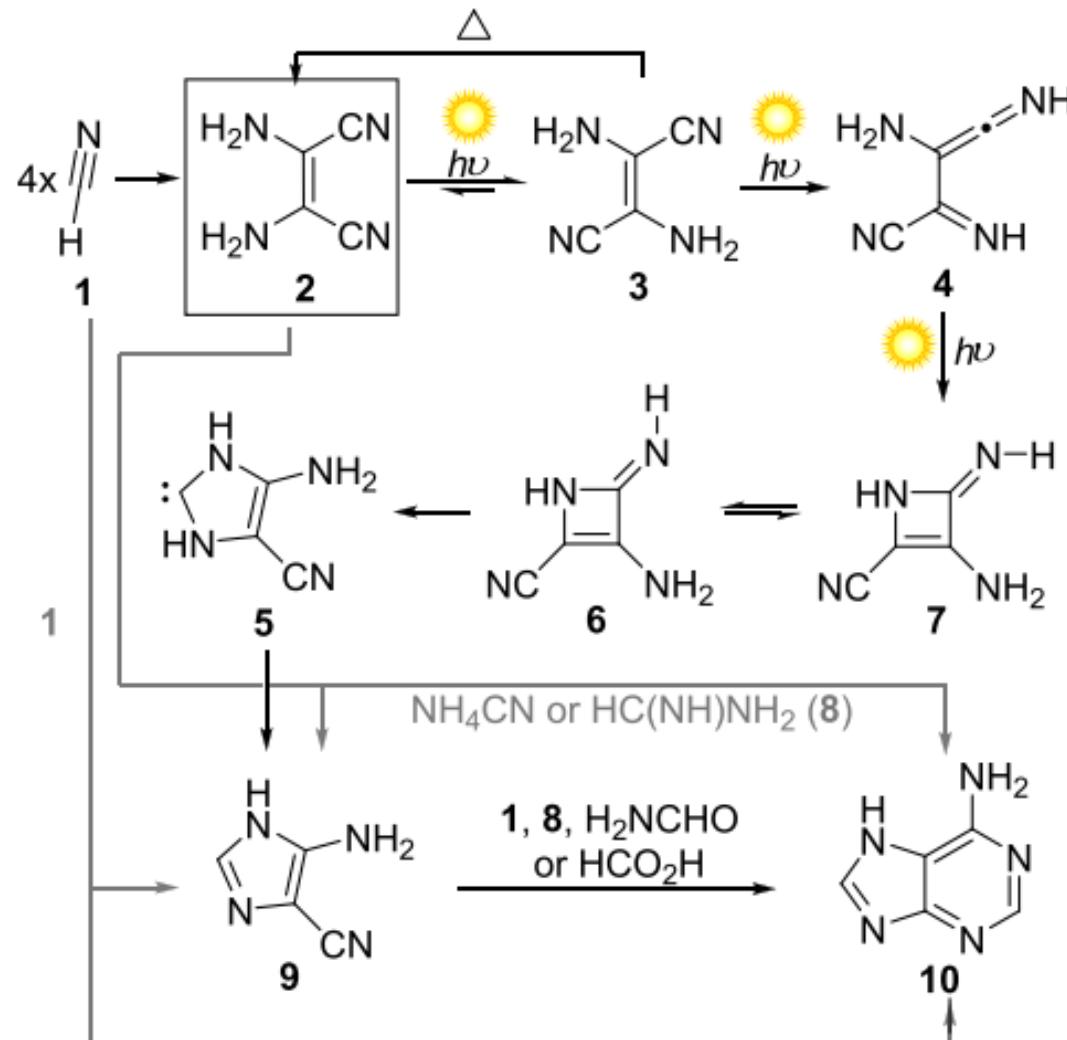
Prebiotic soup



Prebiotic soup

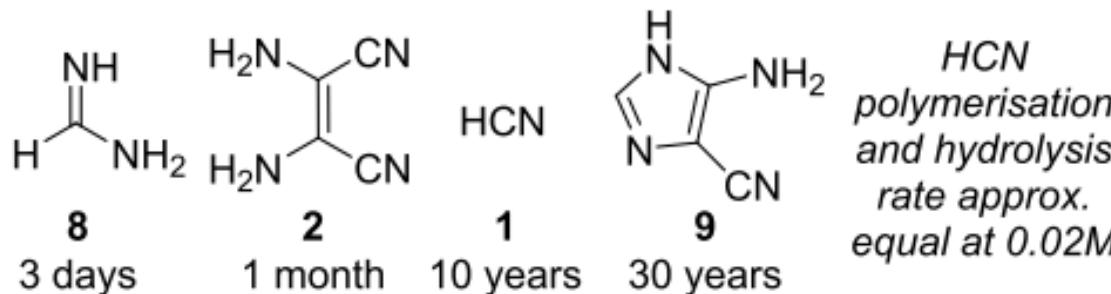


Prebiotic soup



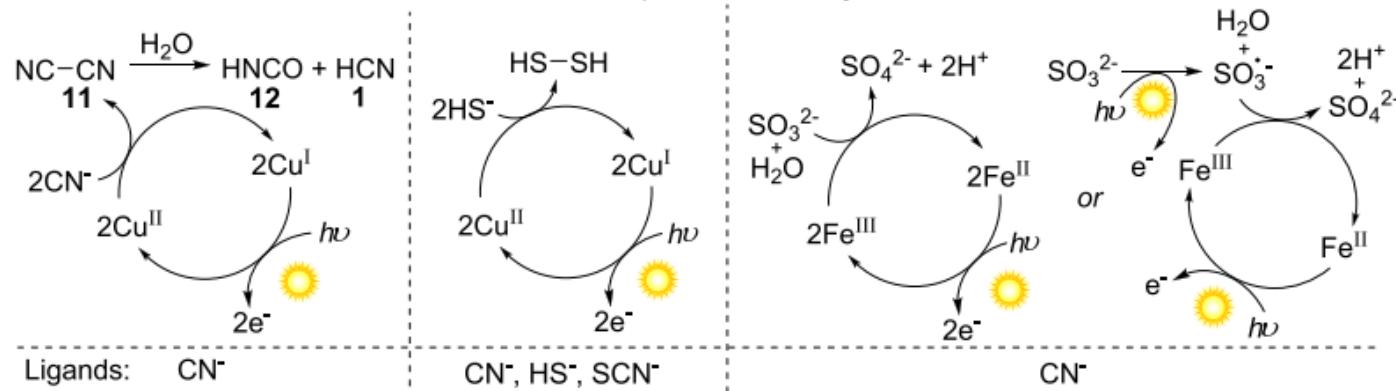
Prebiotic soup

hydrolysis half-times, 30 °C, pH ~8.5



Refinement of geochemical scenario

Improved chemistry



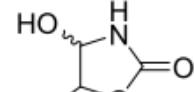
Ligands: CN⁻

CN⁻, HS⁻, SCN⁻

CN⁻

Cu^I / CN⁻

HNCO **12** irreversibly traps **23** and **31** as oxazolidinones



13, R = H or CH₂OH

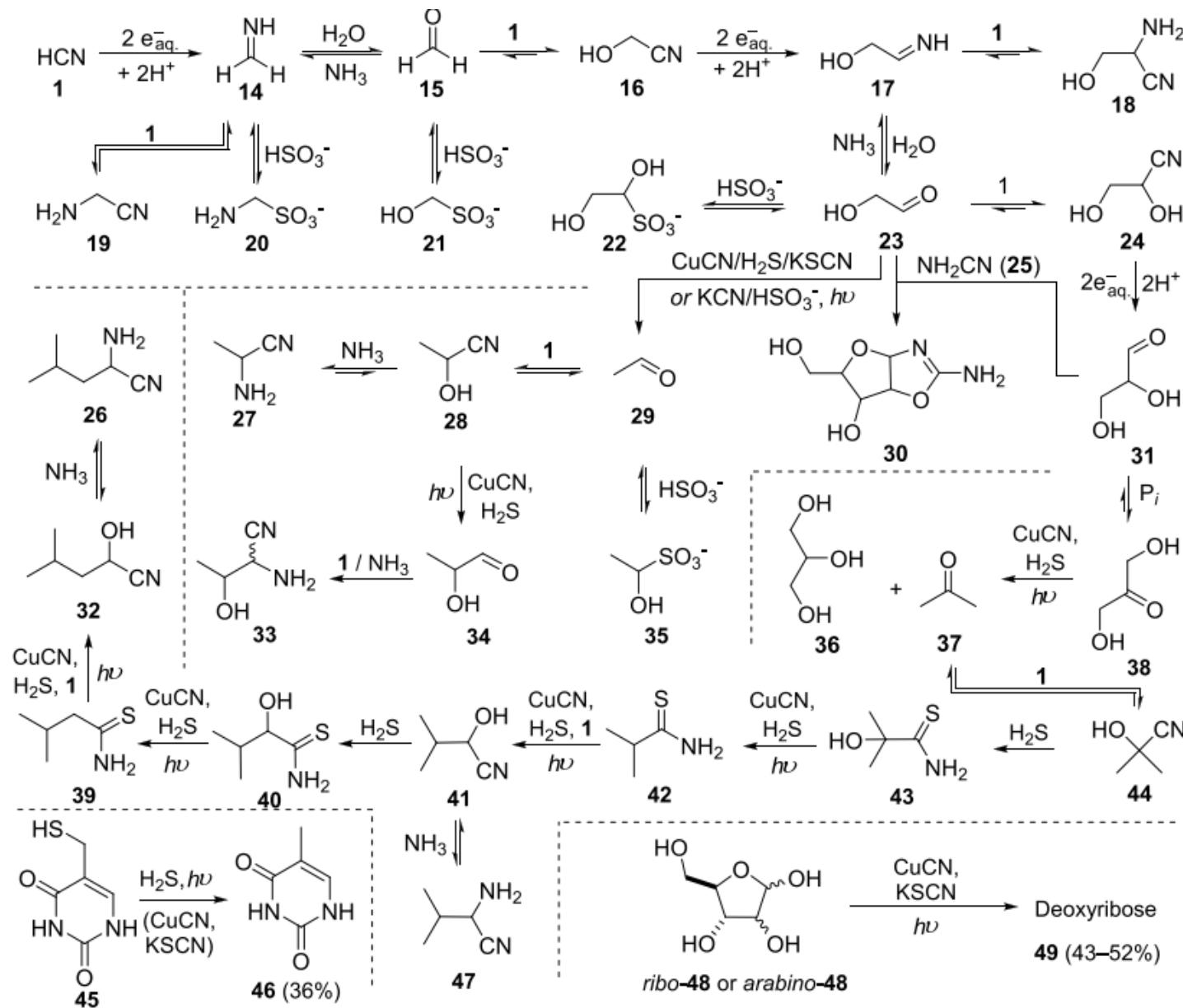
Cu^I / CN⁻ / HS⁻

Difficult to generate reasonable [HS⁻] via equilibrium from atmospheric H₂S
Cu^I of relatively low crust abundance

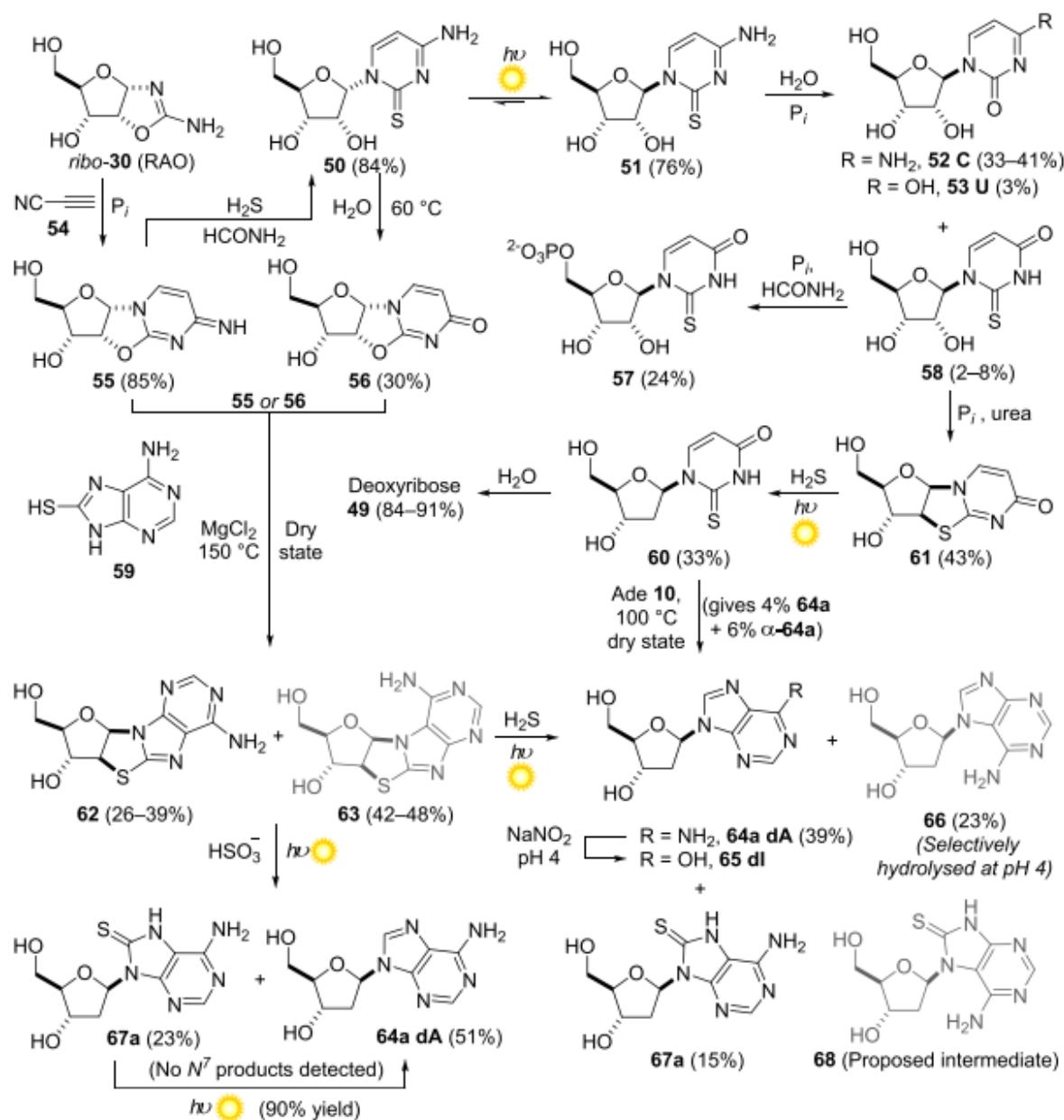
Fe^{II} / CN⁻ / SO₃²⁻

Fe(CN)₆⁴⁻ provides both plausible repository of CN⁻ and photoredox couple
Atmospheric SO₂ easily concentrated in solution globally as SO₃²⁻
Relatively high crust abundance of Fe^{II}

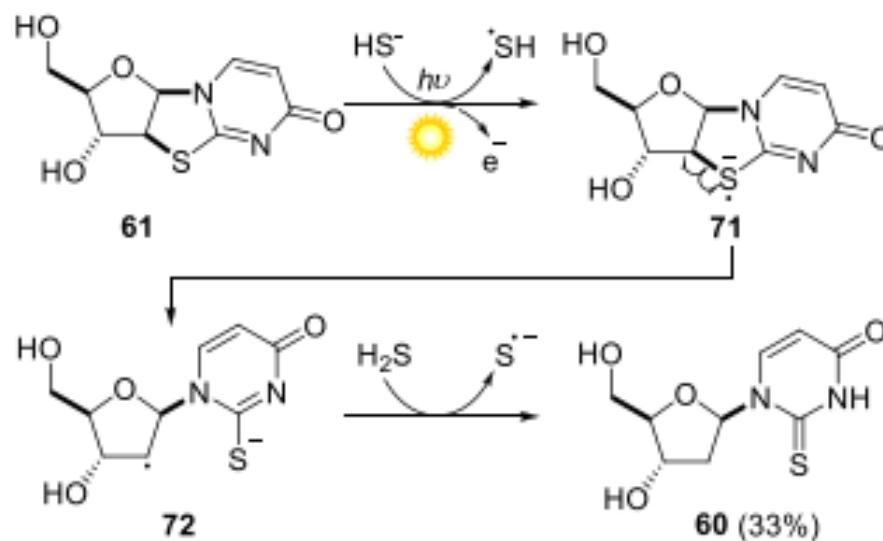
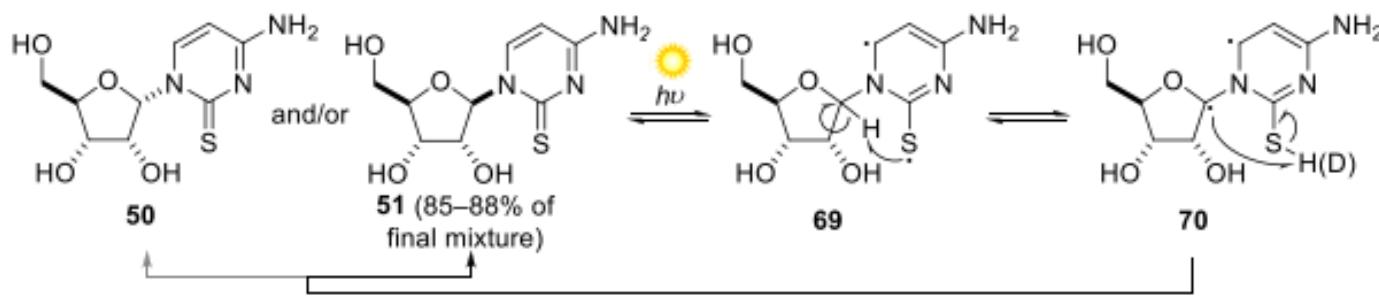
Prebiotic soup



Prebiotic soup

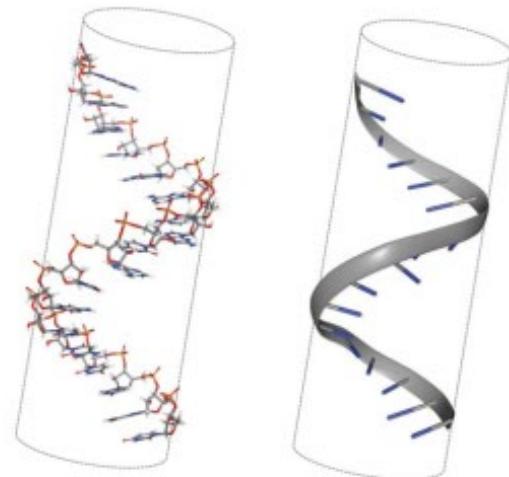


Prebiotic soup

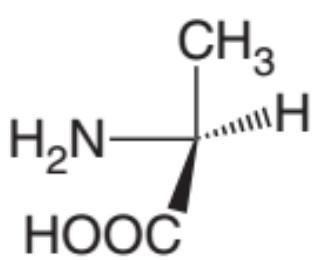




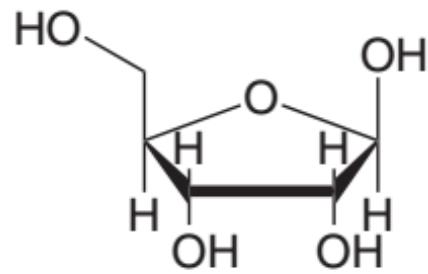
D-RNA (live system)



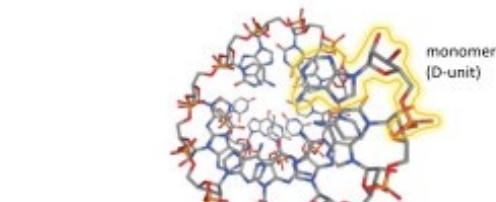
Right-handed helix
(side view)



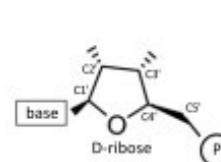
L-amino acids



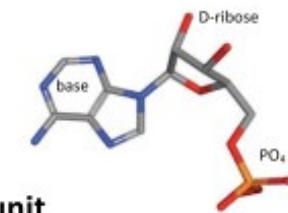
D-amino acids

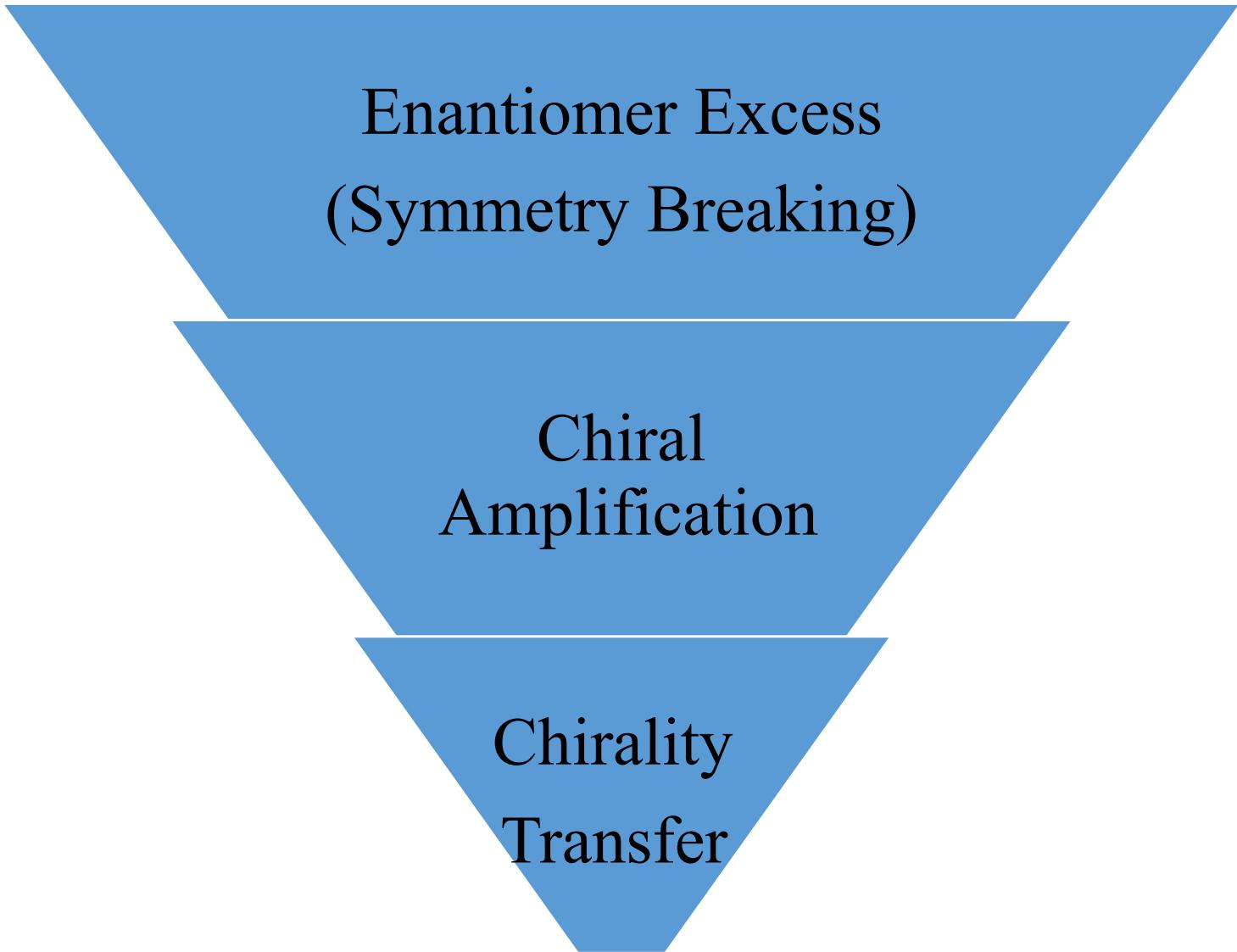


(top view)



D-unit





Enantiomer Excess (Symmetry Breaking)

Chiral Amplification

Chirality Transfer

Symmetry, **2021**, *13*, 2277

Cold Spring Harb Perspect Biol 2019;11:a032540

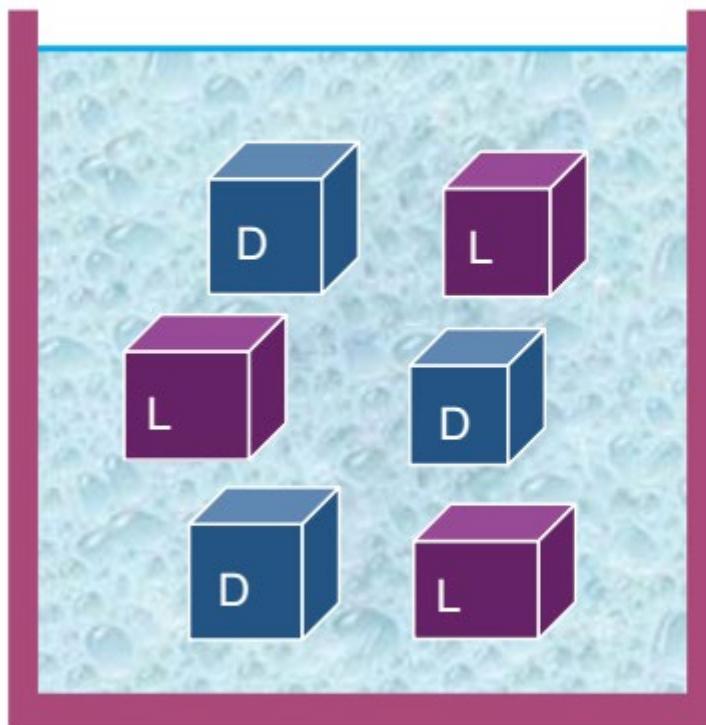
Symmetry Breaking

Enantiomeric excesses in amino acids from the Murchison and Murray meteorites.

Amino acid	Sample				Standard				Confidence (%)	Corrected ee (%)
	L (%)	σ	n	ee (%)	L (%)	σ	n	ee (%)		
<i>Murray meteorite</i>										
2-Amino-2,3-dimethyl-pentanoic acid										
$2S,3S/2R,3R$	50.6	0.7	12	1.2	50.1	0.6	16	0.2	>95	1.0
$2S,3R/2R,3S$	51.0	0.6	12	2.0	49.9	0.6	17	-0.2	>99	2.2
Isovaline	53.6	0.4	20	7.2	50.6	0.3	24	1.2	>99.9	6.0
α -Methylnorvaline	51.1	0.3	23	2.2	50.4	0.3	30	0.8	>99.9	1.4
α -Methylvaline	51.1	0.8	21	2.2	50.6	0.7	17	1.2	>90	1.0
α -Methylnorleucine	51.1	0.5	18	2.2	50.2	0.6	19	0.4	>99.9	1.8
α -Amino-n-butyric acid	50.8	0.5	14	1.6	51.0	0.4	20	2.0	Not sig.	-0.4
Norvaline	50.9	0.5	10	1.8	50.5	0.9	17	1.0	Not sig.	0.8
Alanine	51.0	0.3	21	2.0	50.8	0.3	27	1.6	>95	0.4
Valine	50.6	0.6	14	1.2	50.8	0.4	13	1.6	Not sig.	-0.4
<i>Murchison meteorite</i>										
* 2-Amino-2,3-dimethyl-pentanoic acid										
$2S,3S/2R,3R$	52.6	0.5	5	5.2	48.8	1.9	14	-2.4	>99	7.6
$2S,3R/2R,3S$	54.7	0.6	5	9.4	50.1	0.8	18	0.2	>99.9	9.2
* Isovaline	54.6	0.6	8	9.2	50.4	0.6	15	0.8	>99.9	8.4
* α -Methylnorvaline	51.4	0.4	10	2.8	50.0	0.2	10	0	>99.9	2.8
α -Methylvaline	51.6	0.3	7	3.2	50.2	0.6	10	0.4	>99.9	2.8
α -Methylnorleucine	52.5	1.3	10	5.0	50.3	0.4	13	0.6	>99.9	4.4
* α -Amino-n-butyric acid	50.4	0.2	3	0.8	50.2	0.2	12	0.4	Not sig.	0.4
*Norvaline	50.2	0	3	0.4	50.0	0.2	10	0	Not sig.	0.4
Alanine	50.8	0.4	15	1.6	50.2	0.3	16	0.4	>99.9	1.2
Valine	51.3	0.4	12	2.6	50.2	0.2	13	0.4	>99.9	2.2

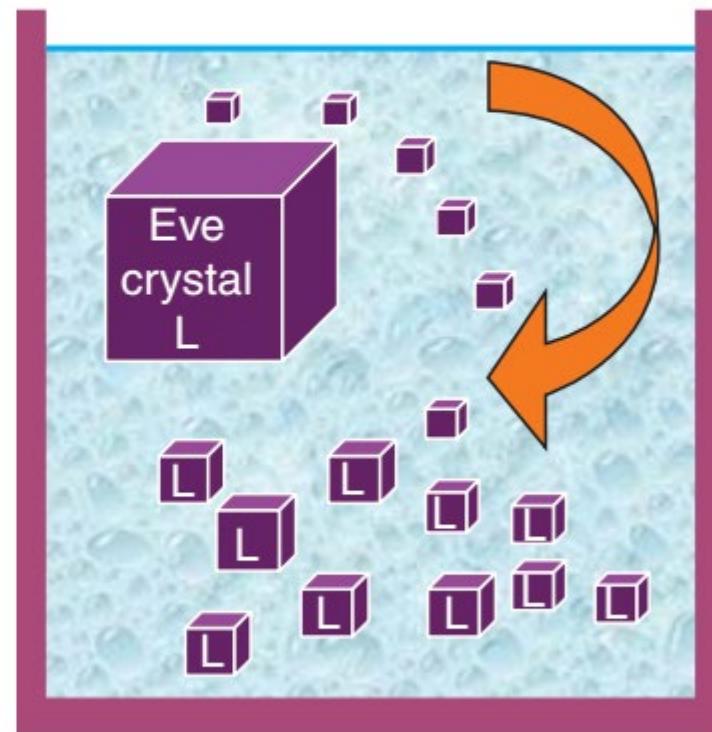
Symmetry Breaking

Slow primary nucleation
of multiple crystals



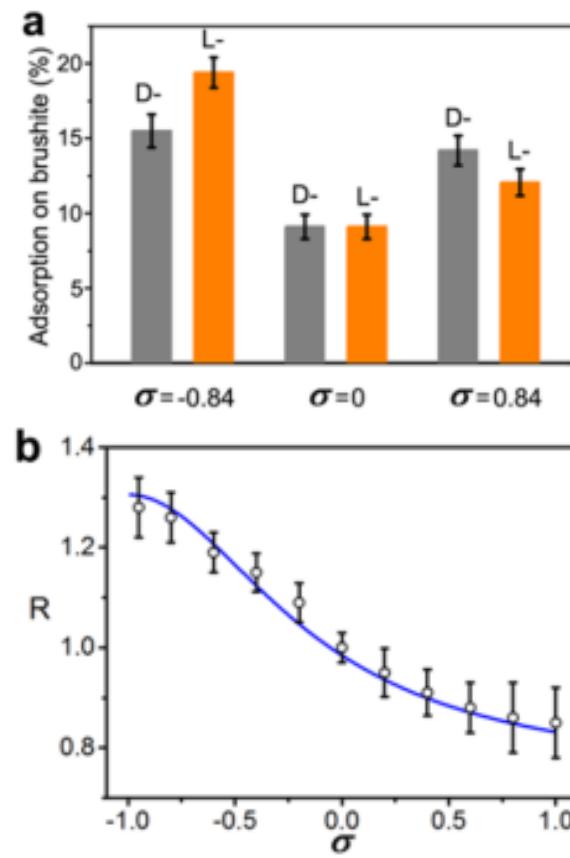
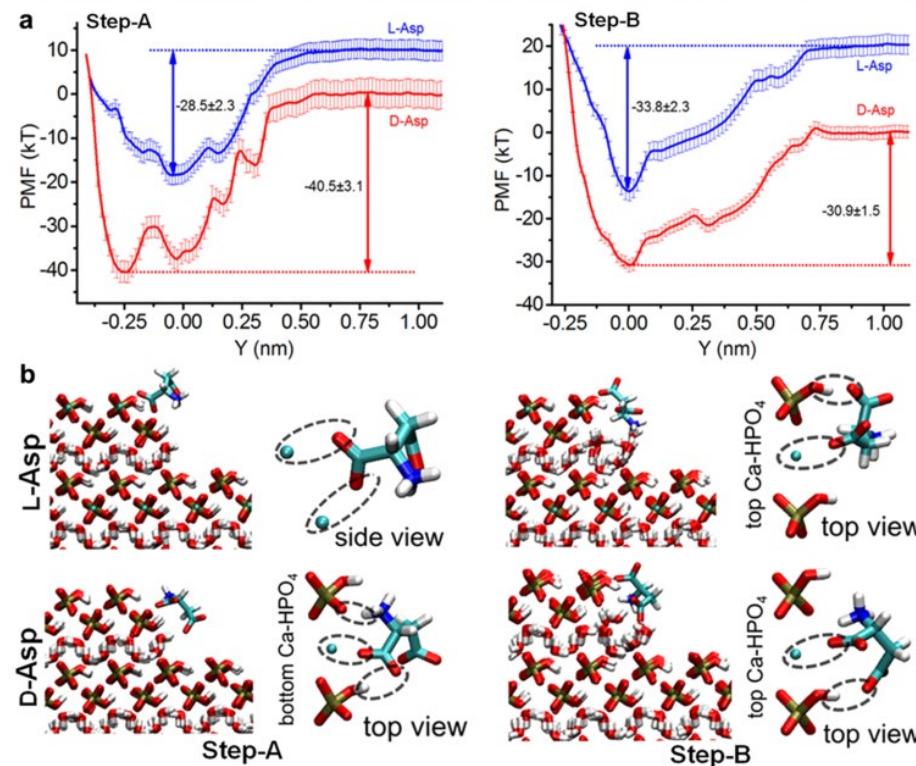
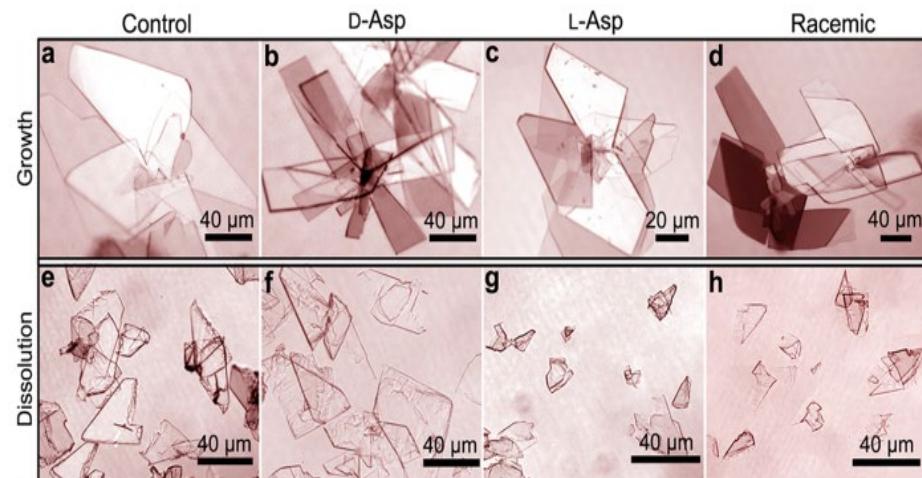
No stirring

Rapid secondary nucleation due to
shear of first crystal formed

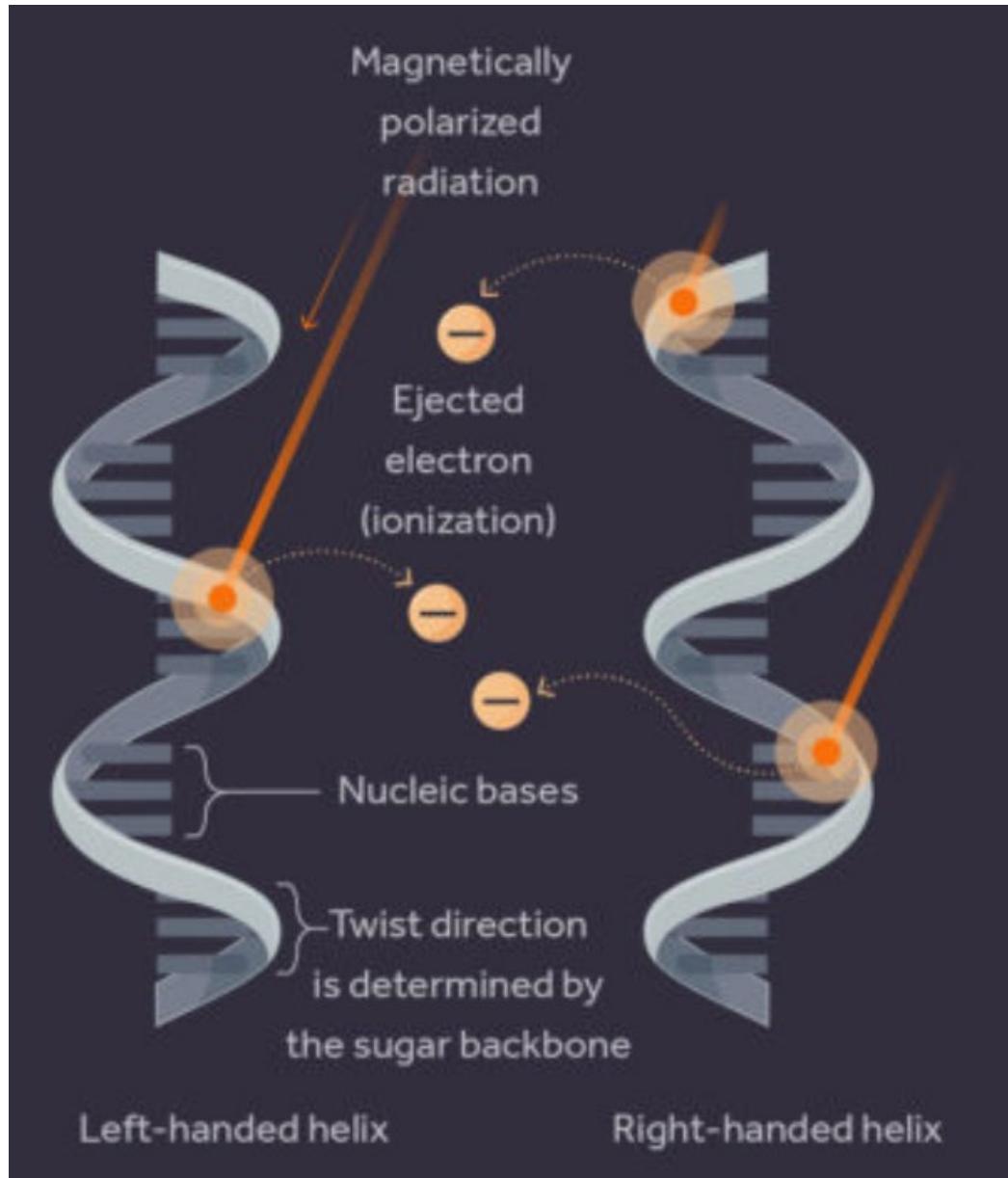


Rapid stirring

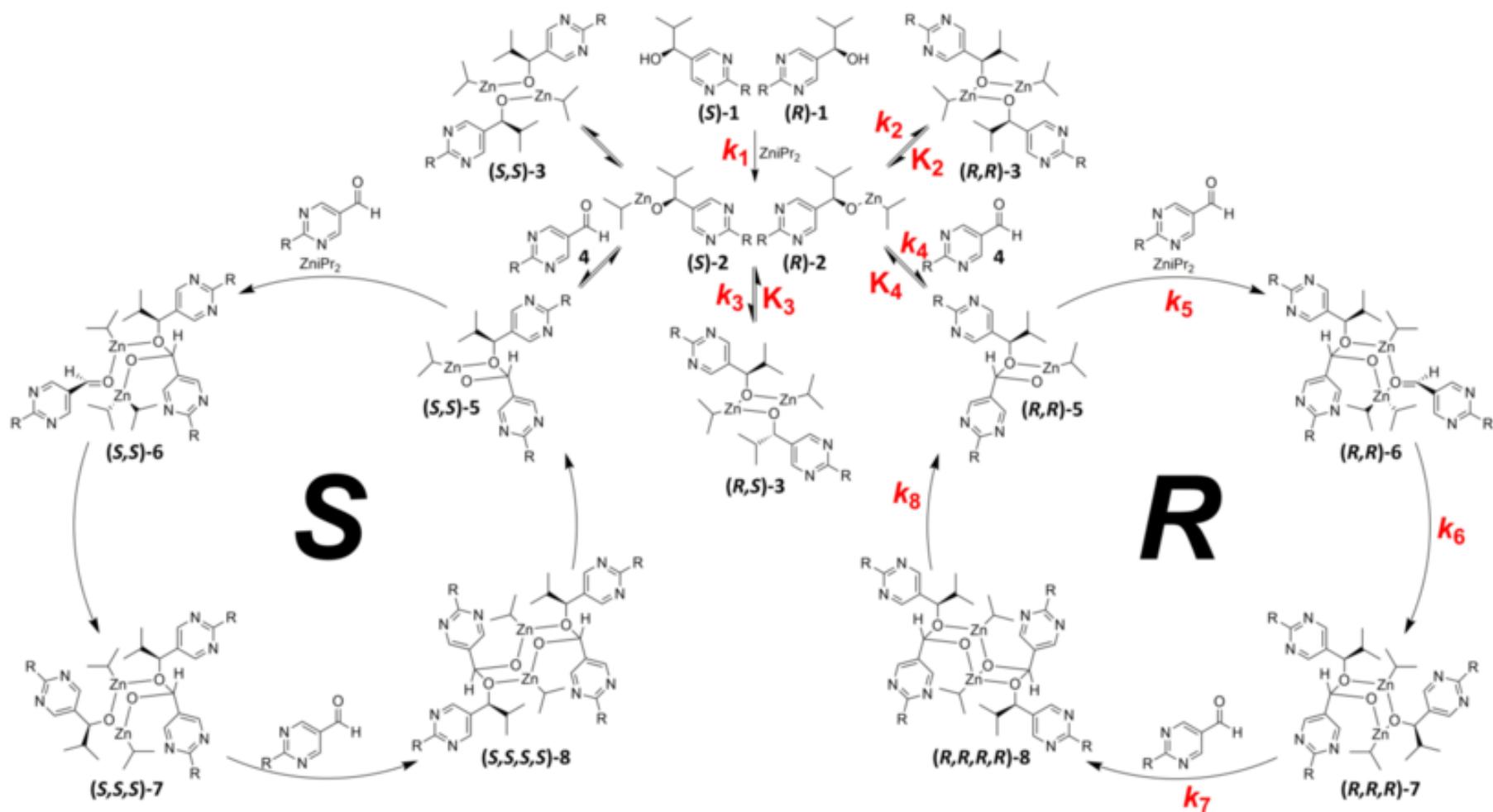
Symmetry Breaking



Symmetry Breaking



Chiral Amplification

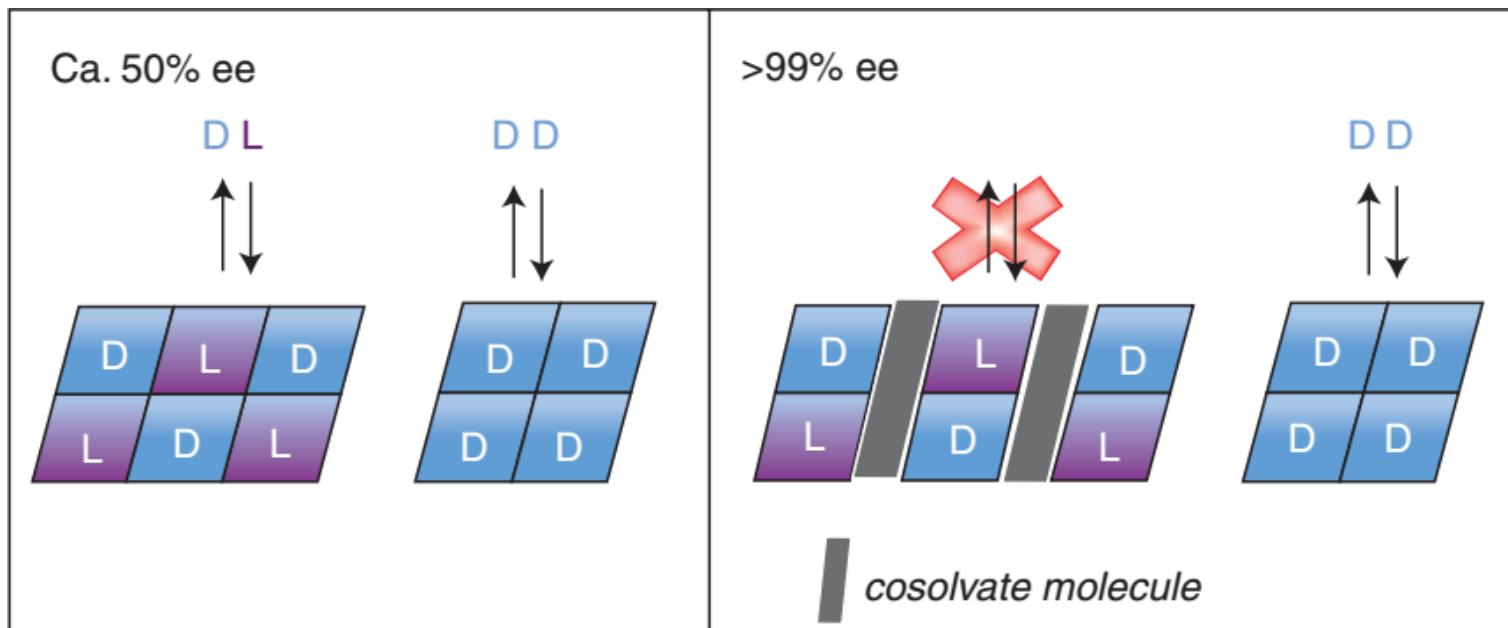


Soai autocatalytic reaction

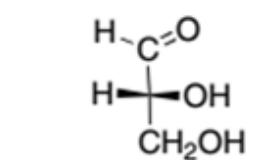
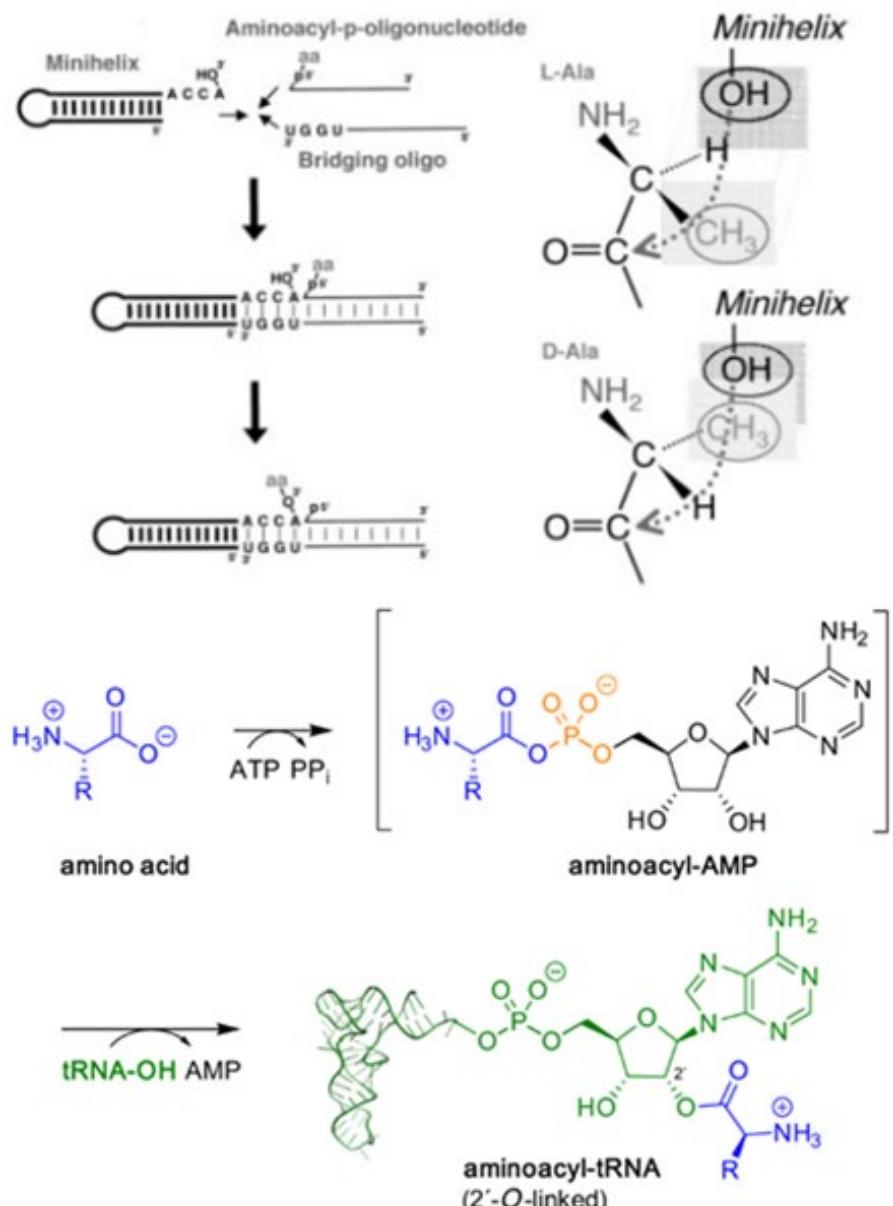
Chiral Amplification

Enantiomeric concentration amplification of phenylalanine after two crystallizations from water

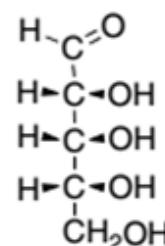
Component	Initial ee, %	Final ee, %
D	10	90.0 ± 3.7
	5	91.7 ± 1.5
	1	87.2 ± 2.0
L	10	88.3 ± 1.1
	5	88.6 ± 0.9
	1	90.9 ± 0.3



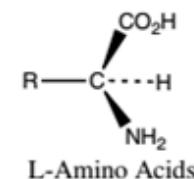
Chirality Transfer



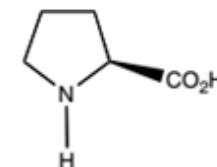
D-Glyceraldehyde



D-Ribose



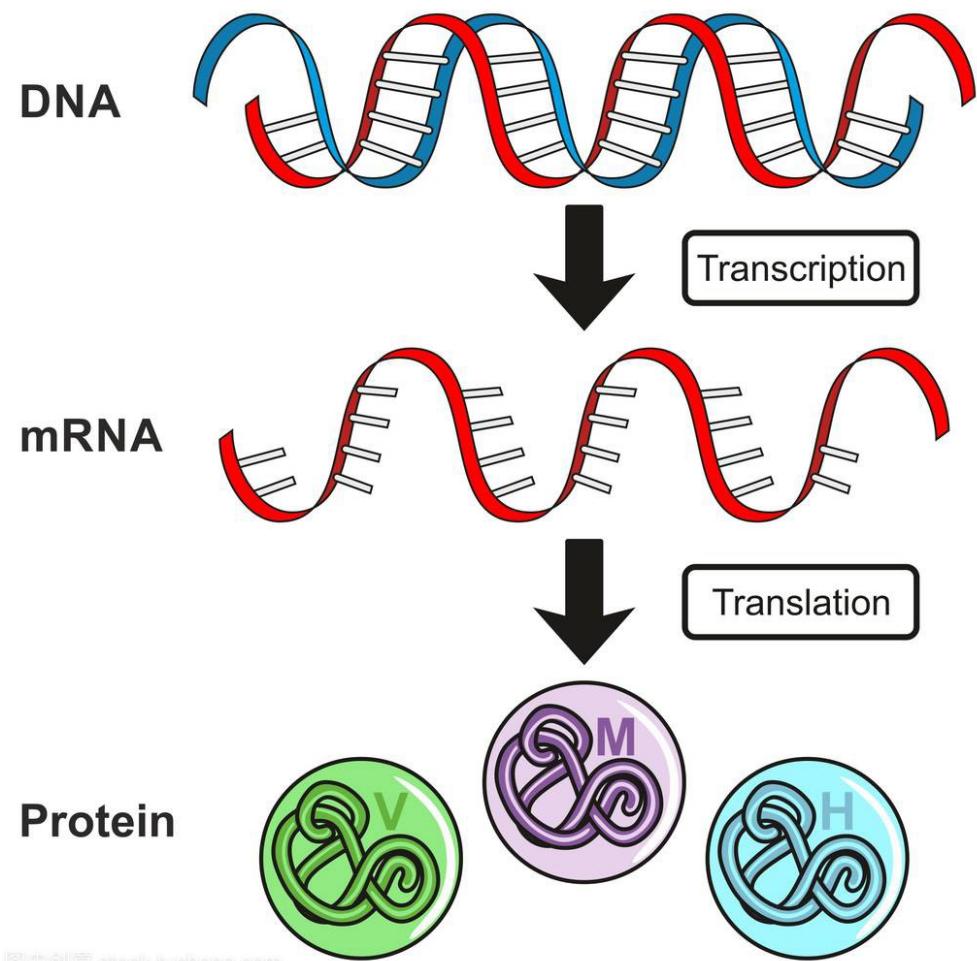
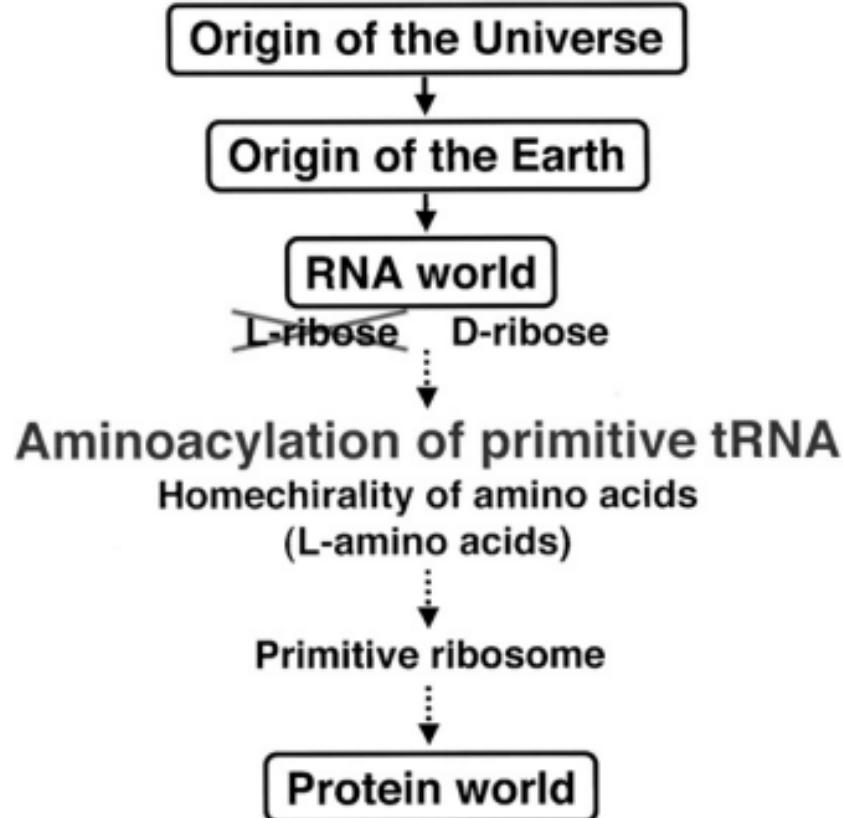
alanine ($\text{R} = \text{CH}_3$); serine ($\text{R} = \text{CH}_2\text{OH}$); valine ($\text{R} = \text{CH}(\text{CH}_3)_2$); leucine ($\text{R} = \text{CH}_2-\text{CH}(\text{CH}_3)_2$); phenylalanine ($\text{R} = \text{CH}_2-\text{C}_6\text{H}_5$); glutamic acid ($\text{R} = \text{CH}_2-\text{CH}_2-\text{CO}_2\text{H}$)



L-Proline

Amino acid	Ratio, D/L
L-serine	50.3/49.7
L-alanine	50.8/49.2
L-phenylalanine	52.2/47.8
L-valine	52.2/47.8
L-leucine	54.4/45.6
L-glutamic acid	60.7/39.3
L-proline	28.9/71.1

Chirality Transfer



图示创意 stock.tuchong.com

Thanks !