

Synergistic photobiocatalysis for enantioselective triple

radical sorting

徐浩文

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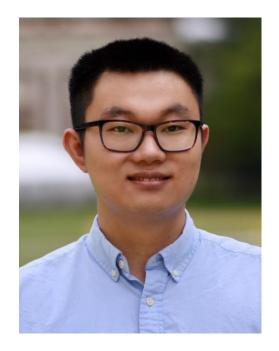
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Author



Dr. Xiaoqiang Huang

State Key Laboratory of Coordination Chemistry, Chemistry and Biomedicine Innovation Center (ChemBIC), Nanjing University

- B.E, July 2013, Peking University
- M.S, July 2015, Peking University (Supervisor: Prof. Ning Jiao)
- PhD, February 2019, Philipps-Universität Marburg (Supervisor: Prof. Eric Meggers)
- Post-Doctor training, March 2021, UIUC (Supervisor:Huimin Zhao)
- Research interest: Repurposing Natural Enzymes with Light and Engineering

Author



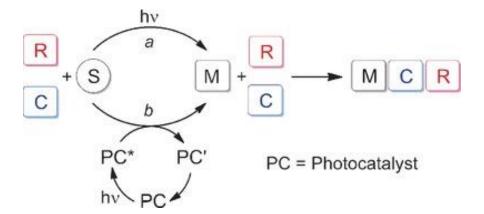
Dr. Binju Wang

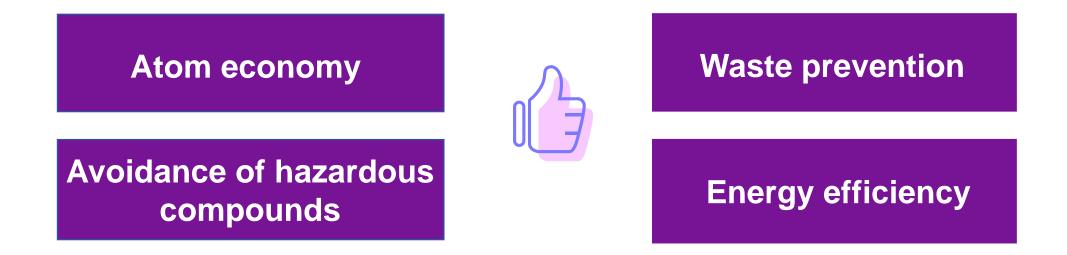
College of Chemisty and Chemical Engineering, Xiamen University

- B.E, July 2007, Harbing Institute of Technology
- PhD, July 2012, Xiamen University
- Post-Doctor training, September 2016, The Hebrew University of Jerusalem
- Research interest: Multiscale theoretical simulation of chemical reactions in aqueous solutions and protein environments

Multicomponent reactions

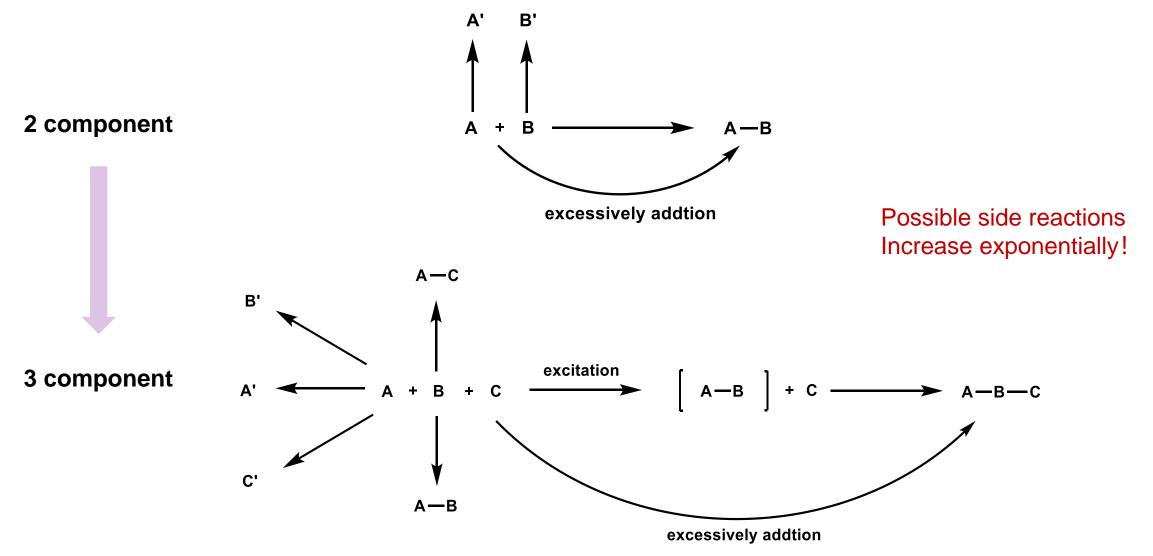
- More than two substrates react together in a multistep, one-pot process to form a single product, with most of the atoms of the starting materials incorporated
- Rapid achievement of high levels of complexity and diversity
- Practical and time-saving operations



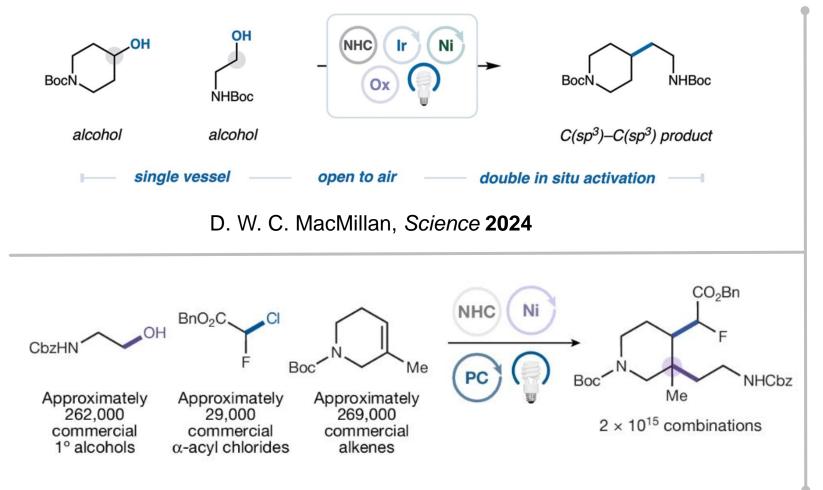


S. Protti, A. Basso et al., Angewandte Chemie International Edition 2016, 55, 15476-15484.

Challenges of radical multicomponent reactions: chemoselectivity

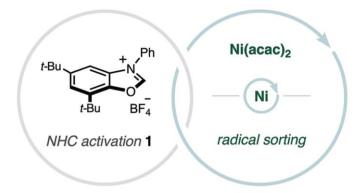


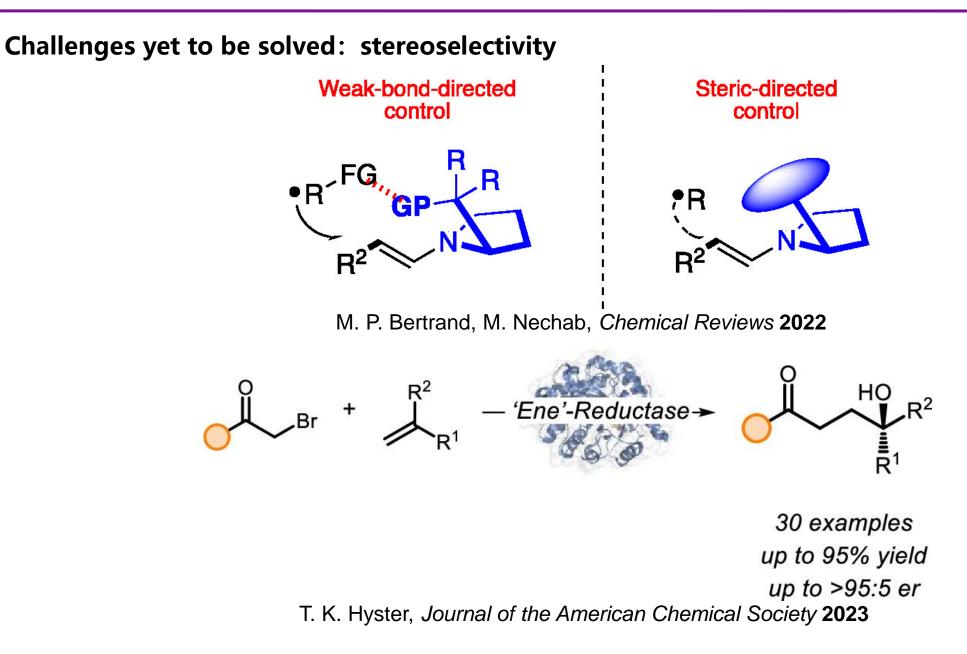
Chemical catalytic radical sorting



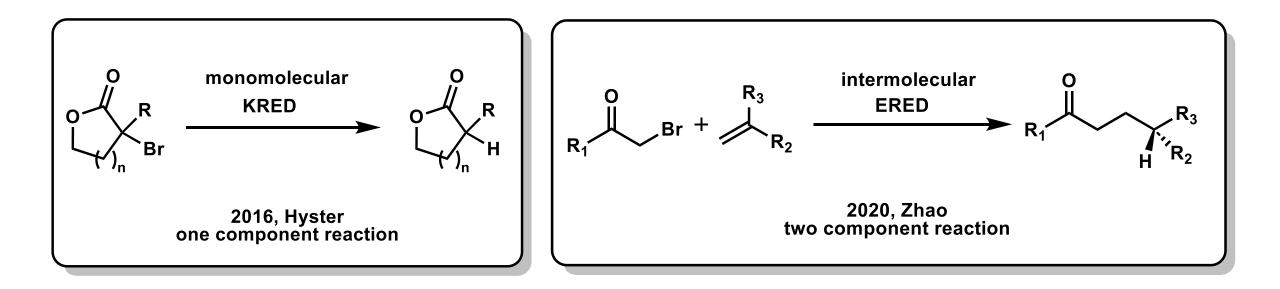
D. W. C. MacMillan, *Nature* **2024**

Chemical catalytic radical sorting has emerged as an enabling strategy for a variety of useful reactions





Milestones in new-to-nature photobiocatalysis





How to use one protein to regulate the orderly transformation of three substrates?

Target

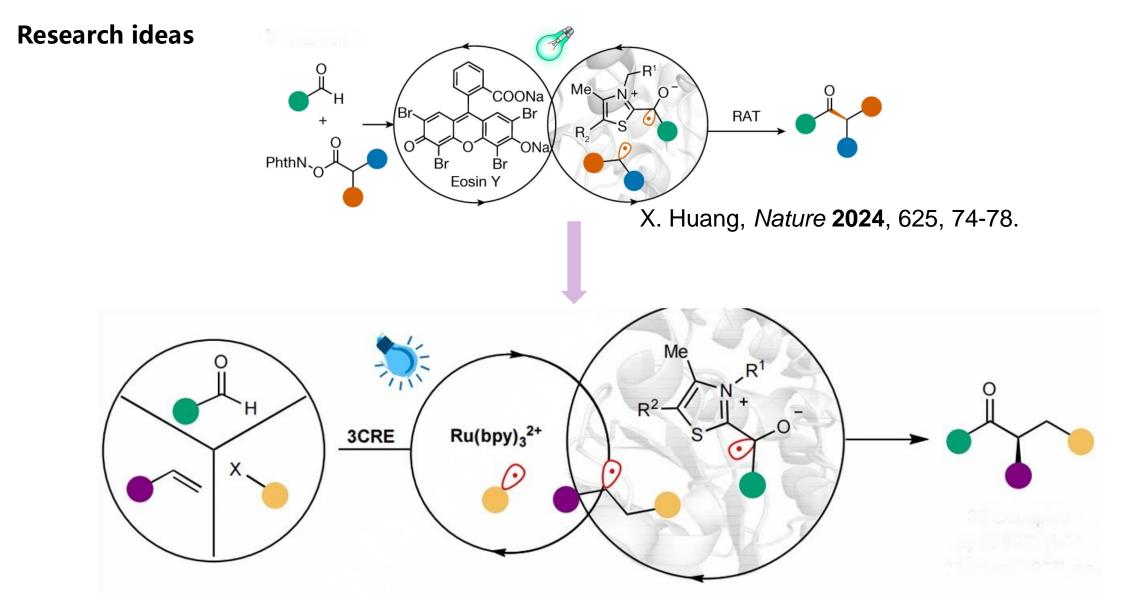


 Develop a new photobiocatalytic method to achieve a triple radical coupling reaction **••**

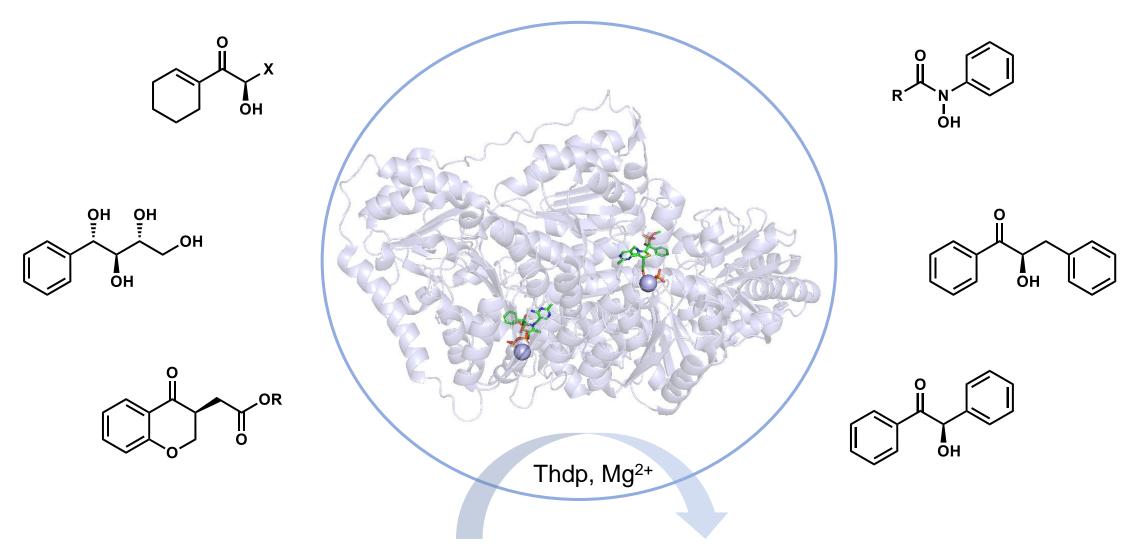
• Enzymes evolve for specific functions

Restricted confines of the active site

Reactions involving multiple substrates/radicals
 are inherently difficult

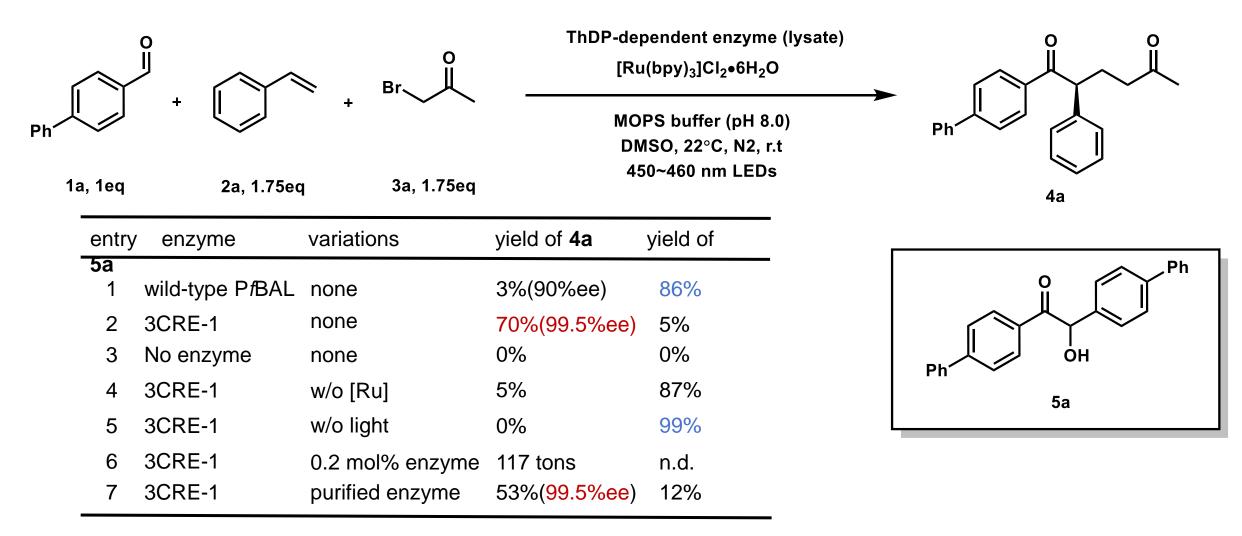


PfBAL

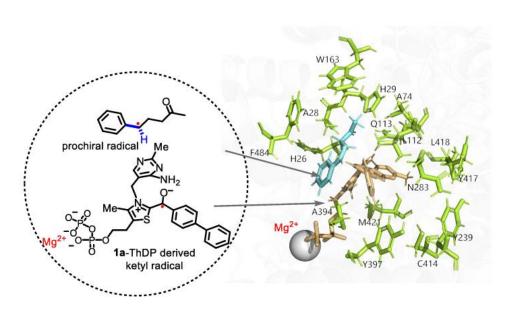


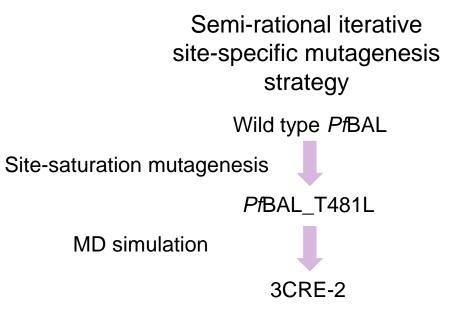
D. C. Howell, F. Chen, Green Synthesis and Catalysis 2022

Standard conditions and control experiment



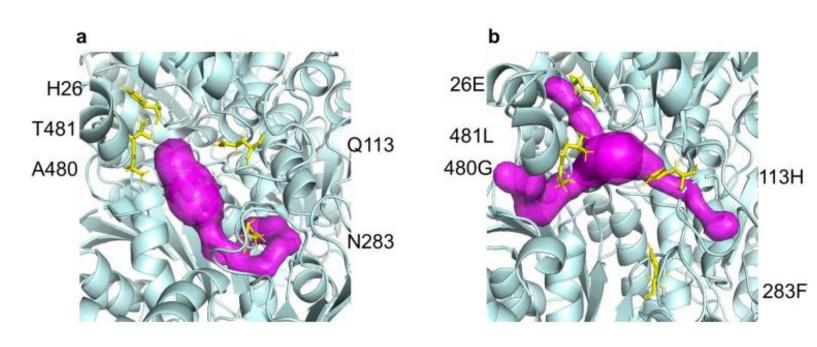
Reaction development and engineering





entry	conditions	4a (yield, ee)	5a (yield, ee)	5b (yield)	5c (yield)
1	Wild type PfBAL	3% y, 90% ee	86% y, 98% ee	9% у	1.6% y
2	PfBAL_T481L	10% y, 95% ee	67% y, 98% ee	21% у	1% y
3	<i>Pf</i> BAL_T481L-A480G	18% y, 99% ee	67% y, 98% ee	14% y	1% y
4	<i>Pf</i> BAL_T481L-A480G-Q113H	48% y, 98% ee	14% y, 98% ee	38% у	0.3% y
5	<i>Pf</i> BAL_T481L-A480G-Q113H-N283F (3CRE-2)	66% y, 98% ee	5% y, 94% ee	29% у	1.4% y
6	<i>Pf</i> BAL_T481L-A480G-Q113H-N283F-H26E (3CRE-1)	70% y, 99.5% ee	5% y, 92% ee	23% у	2.3% y

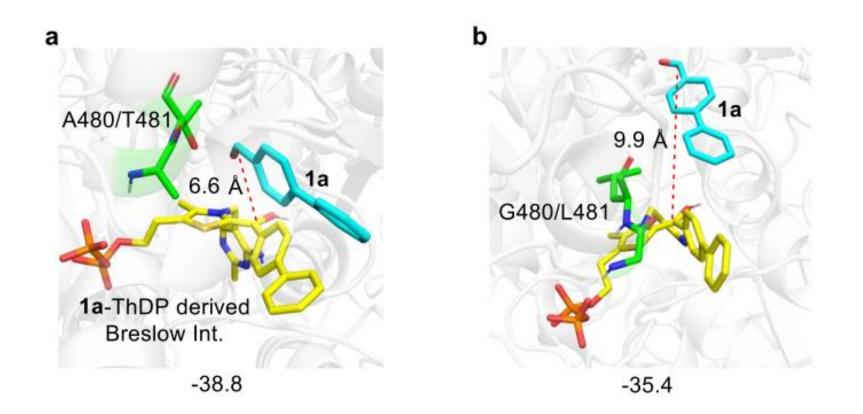
MD revealed the mechanism of action of several key mutation sites



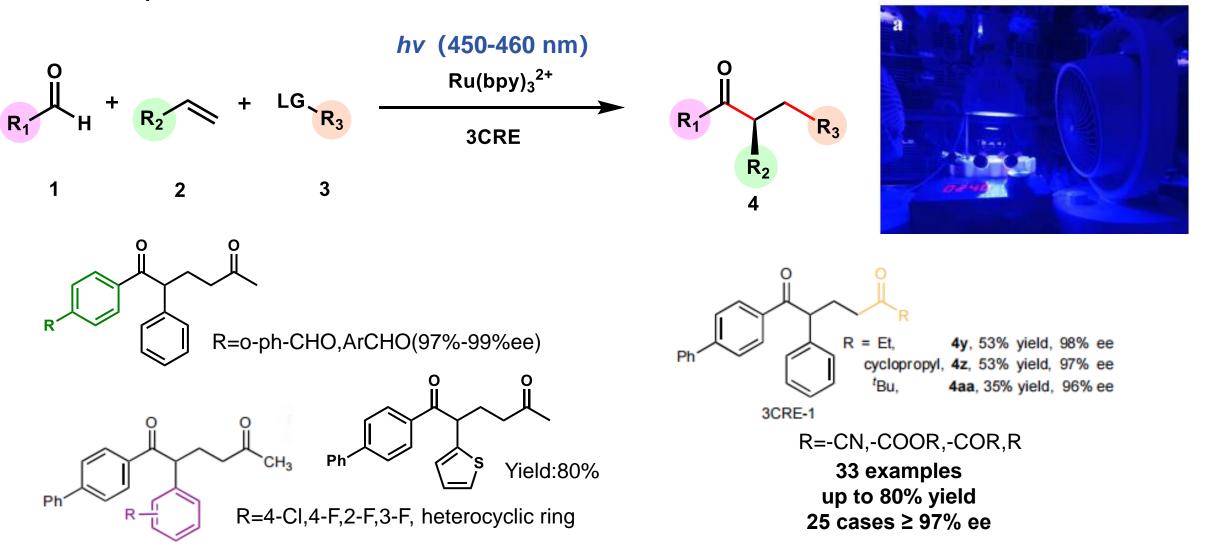
The tunnels of the wild-type PfBAL (a) and the five-site mutated 3CRE-1 (b) CAVER Web 1.0¹ PfBAL: PDB code: 3D7K

- L481 and G480 double mutant significantly enhanced enantioselectivity
- H113 mutant greatly improves the yield and decreases the native benzoin condensation
- The volume of tunnels has expanded from 1835 Å³ for the wild type to 2210 Å³ for the 3CRE-

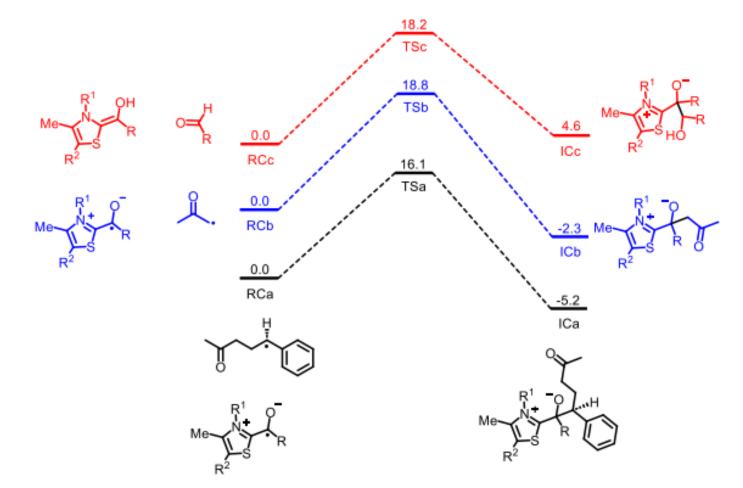
MD simulated structures of the key intermediate for benzoin condensation enabled by (a) wild-type PfBAL, and (b) 3CRE-1



Substrate scope

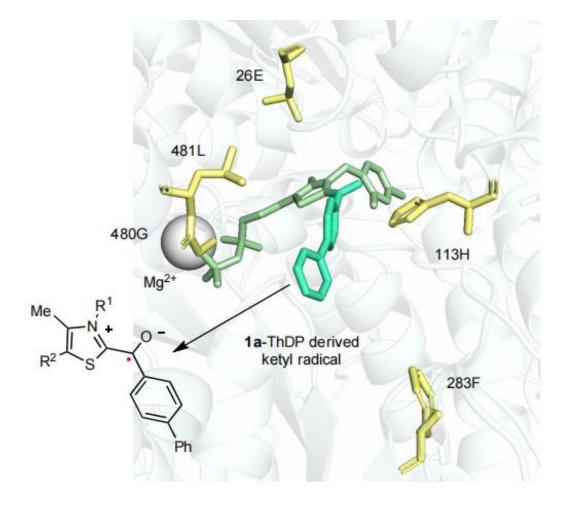


Side reactions vs main reaction



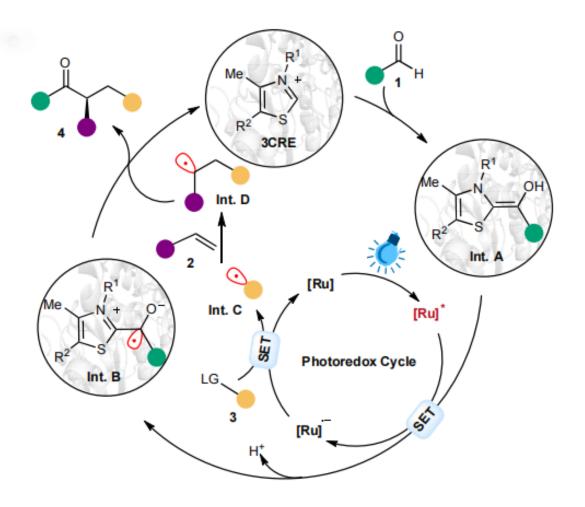
Geometry optimization、IRC: B3LYP-D3/def2-SVP Single point energy: B3LYP-D3/def2-TZVP

Mutated sites view



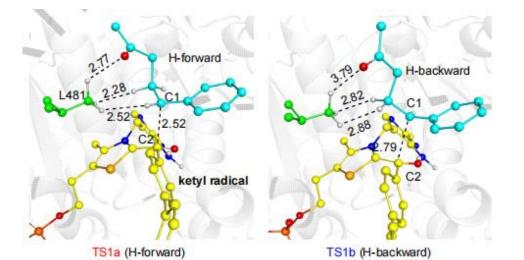
QM (B3LYP-D3/B2)/MM

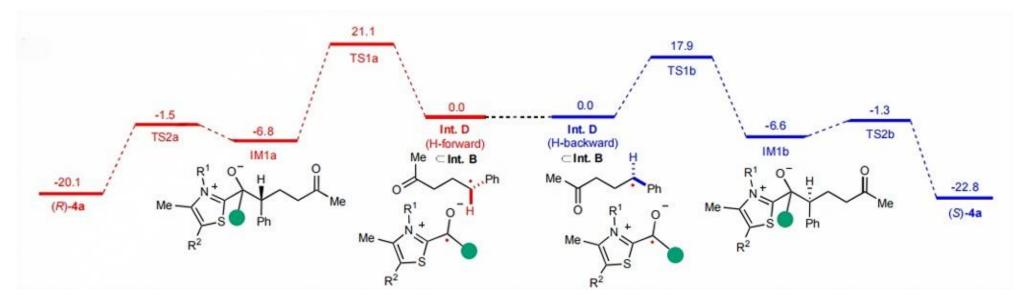
Proposed mechanism



- a) fomation of Breslow intermediate
- b) [Ru] complex oxidate int.A and
 [Ru]-reduce the bromo-substrate to generate the radical(Int.C)
- c) Int.C add to alkene to generate int.D
- d) Int. B and Int. D undergo
 stereoselective radical coupling
 and regenerating the enzyme

The steric hindrance effect is the key to achieving excellent enantioselective control





Conclusion

- Achieving chemo-/enantio-selective enzymatic triple radical sorting scaled to preparative dimensions (gram scale)
- System with excellent substrate tolerance, outstanding stereochemical control and mild reaction conditions, which is not easily attainable with chemo-NHC
- A synergistic dual photoredox and biocatalytic cycle mechanism is proposed

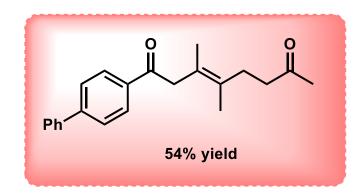
Prospect

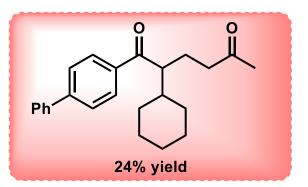
- Provide a broad playground for exploring new-to-nature biocatalysis system
- Evolved enzymes containing NHC cofactors have more potential for development.

Discussion

Limitations

- Show low yield when involving unactivated alkene
- Excessive amounts of alkene and bromocarbonyl compound (1.75 equivalents) were used in the reaction (poor atom economy)
- Calculation: theory method (B3LYP-D3/def2-TZVP) may need optimization
- Lack of direct experimental evidence regarding the binding site of the [Ru] photocatalyst on the enzyme





Driving innovation in chemical synthesis methods

Expanding the boundaries of biocatalytic functions

Deepening the understanding of the synergy mechanism of ThDP-dependent biocatalysis Facilitating the development of evolved enzymes





Thanks for listening!



